

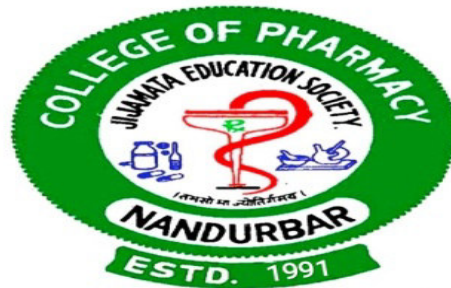
Liquid Orals

By

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CONTENTS:-

- Definition of liquid orals
- Types of liquid orals
- Suspensions
- Emulsion
- Solutions

DEFINITION:-

Liquid orals are the homogeneous liquid preparations containing one or more active ingredients with or without additives dissolved in a suitable vehicle, meant for oral administration.



TYPES OF LIQUID ORALS:-

➤ SUSPENSIONS

➤ EMULSIONS

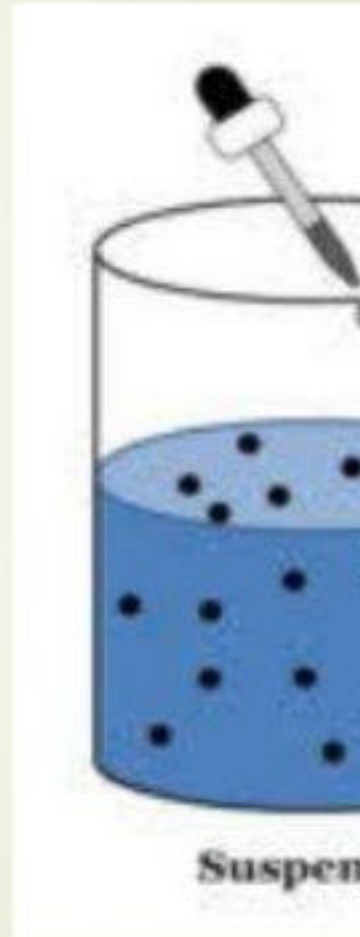
➤ SOLUTIONS

SUSPENSIONS



DEFINITION:-

Suspensions are the biphasic liquid dosage form of medicament in which the finely divided solid particles are suspended or dispersed in a liquid or semisolid vehicle with the help of suspending agent. The solid particle is the 'dispersed phase' or 'discontinuous phase' whereas the liquid vehicle is the 'continuous phase'.



▶ ADVANTAGES:-

- ▶ Can improve chemical stability of certain drugs.
- ▶ Higher rate of bioavailability, as order of bioavailability is:-

Solution > Suspension > Capsules > Compressed tablets

DISADVANTAGES:-

- Physical stability, sedimentation and compaction
- Bulky, handling require care.
- Uniform drug delivery cannot be achieved sometimes.

IDEAL PROPERTIES OF SUSPENSIONS

1. The dispersed particles should not settle readily and the settled particles should redisperse immediately on shaking.
2. The particles shouldn't form a cake on settling.
3. The viscosity should be such that the preparation can be easily poured.
4. It should be chemically stable.
5. Suspensions for internal use must be palatable and suspension for external use must be free from gritty particles.

TYPES OF SUSPENSIONS:-

Depending upon particle nature/dispersibility of particle nature the suspensions are of two types:-

1. Flocculated suspensions
2. Non-flocculated/deflocculated suspensions.

FLOCCULATED SUSPENSIONS:-

Suspension in which particles are weakly bonded, settle rapidly, do not form a cake and are easily resuspended with a minimum of agitation.



DEFLOCCULATED SUSPENSIONS:-

Suspension in which particles settle slowly and eventually form a sediment in which aggregation occurs with the resultant formation of a hard cake which is difficult to resuspend.



Differences between flocculated and deflocculated suspension

Flocculated	Non-flocculated
<p> Particles form loose aggregates and form a network like structure Rate of sedimentation is high Sediment is rapidly formed Sediment is loosely packed and doesn't form a hard cake Sediment is easy to redisperse Suspension is not pleasing in appearance The floccules stick to the sides of the bottle </p>	<ol style="list-style-type: none"> 1. Particles exist as separate entities 2. Rate of sedimentation is slow 3. Sediment is slowly formed 4. Sediment is very closely packed and a hard cake is formed 5. Sediment is difficult to redisperse 6. Suspension is pleasing in appearance 7. They don't stick to the sides of the bottle

FORMULATION OF SUSPENSIONS:-

1. Flocculating agents.
2. Suspending agents/thickening agents.
3. Wetting agents.
4. Dispersing agents.
5. Preservatives.
6. Organoleptic additives.

PREPARATION OF SUSPENSION

Step 1:

Suspensions are prepared by grinding the insoluble materials in the mortar To a smooth paste with a vehicle containing the wetting agent.



Step 2:

Insoluble ingredients are dissolved in same portion of the vehicle and added to the smooth paste to get slurry.



Step 3:

The slurry is transferred to a graduated cylinder, the mortar is rinsed with successive portions of vehicle.



Step 4: Decide whether the solids are

- Suspended in a structured vehicle
- Flocculated
- Flocculated and then suspended

Add the vehicle containing the suspending agent (or) flocculating agent



Step-5:

Make up the dispersion to the final volume .

Thus suspension is prepared.

STABILITY OF SUSPENSIONS:-

A stable suspension can be redispersed homogeneously throughout its shelf life. The more stable pharmaceutical suspensions are flocculated i.e., the suspended particles are bonded together physically to form a loose cake.

EVALUATION OF SUSPENSION STABILITY:-

The following are commonly used for evaluating the physical stability of suspensions:-

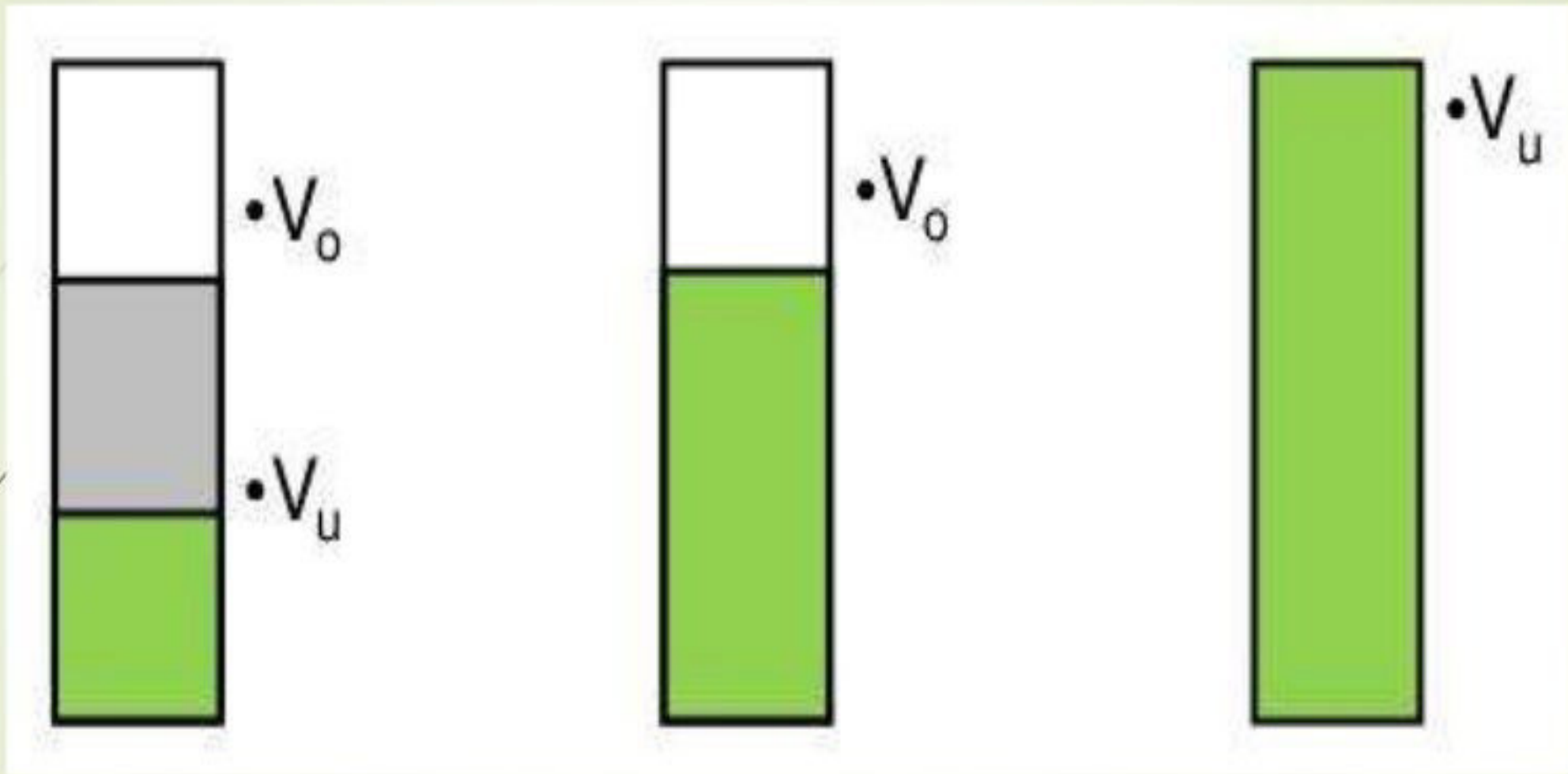
1. Sedimentation method.
2. Rheological method.
3. Electrokinetic method.
4. Micromeritic method.

1. SEDIMENTATION METHOD:-

It is determined by keeping a measured volume of suspension in a graduated cylinder in an undisturbed position for a definite period of time, the ultimate volume (V_0) and the initial volume (V_U) of the sediment is to be noted.

Sedimentation volume is a ratio of the ultimate volume of sediment (V_0) to the original volume of the sediment (V_U) before settling.

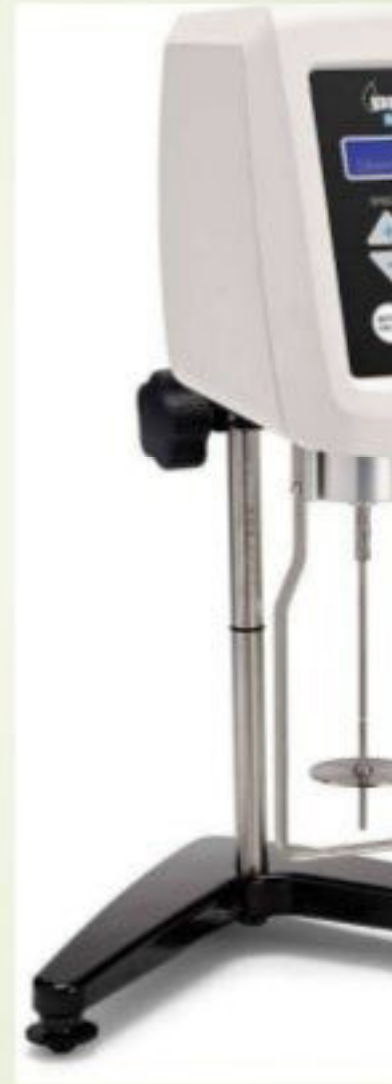
$$\text{Sedimentation volume } F = V_0/V_U$$



Sedimentation rate of a suspension

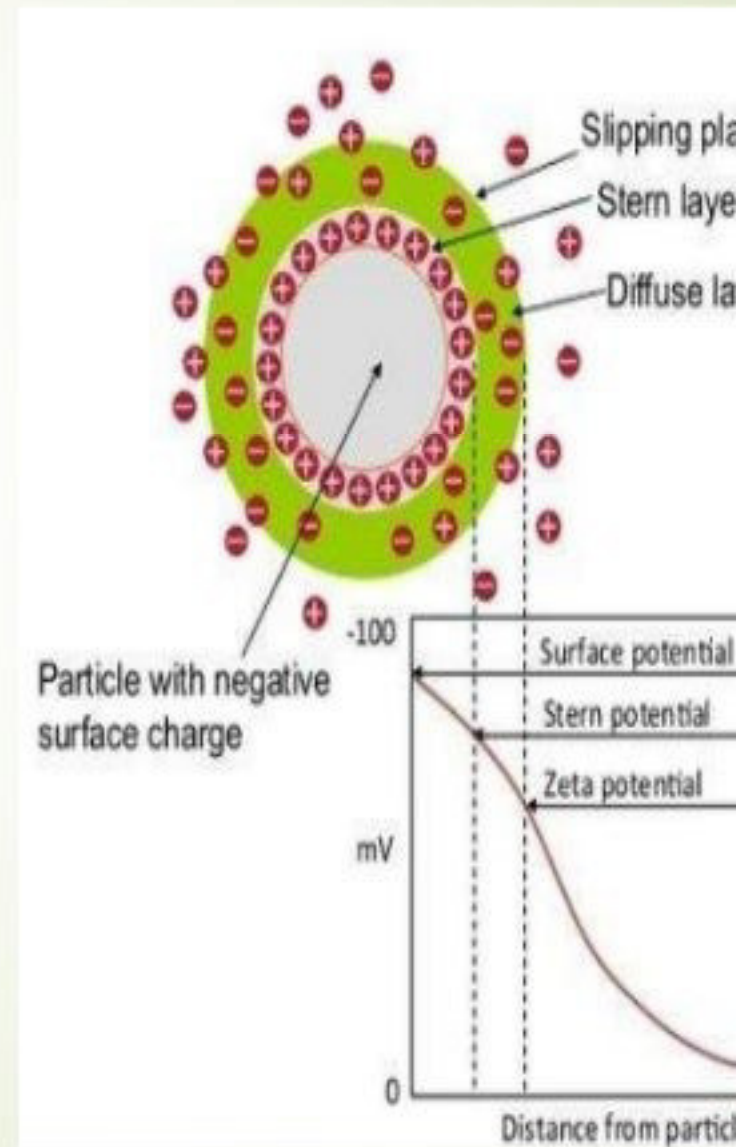
2.RHEOLOGICAL METHOD:-

- It provides information about settling behaviour.
- The arrangement of the vehicle and the particle structural features.
- Brookfield viscometer is used to study the viscosity of the suspension.If viscosity of the suspension increases,the stability of the suspension increases.



3. ELECTROKINETIC METHOD:-

The determination of surface electric charge or zeta potential is helpful to find out the stability of suspension. Zeta potential can be calculated from the migration of particle measured by the electrophoretic method.



4.MICROMERITIC METHOD:-

The stability of suspension depends on the particle size of the disperse phase. The size of the particle in a suspension may grow and ultimately leads to the formation of clumps or caking. So, any change in particle size distribution with reference to time gives a stable suspension. The particle size can be studied by microscopy or coulter-counter method.

EMULSIONS

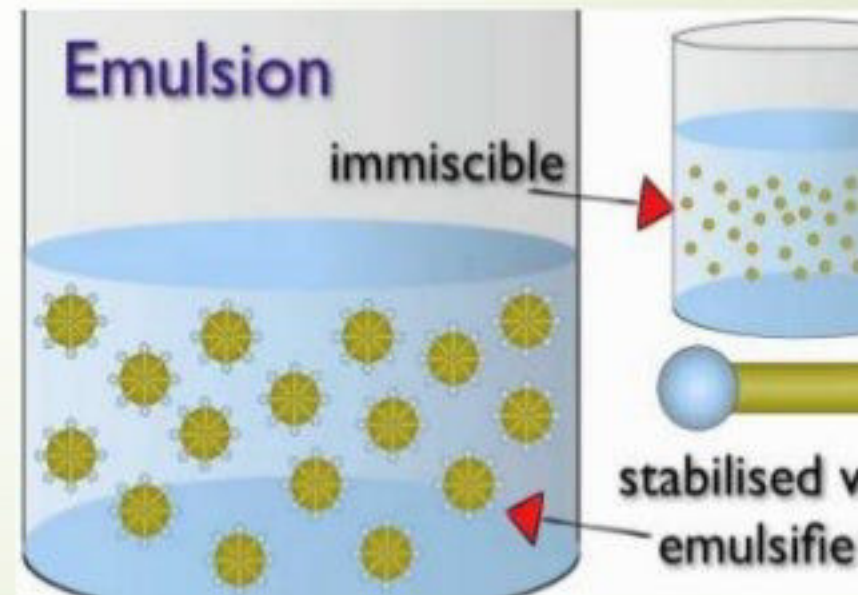


Cream: Oil-in-Water

Butter: Water-in-Oil

DEFINITION:-

An emulsion is defined as a dibasic or heterogenous liquid preparation immiscible liquids which is dispersed as a minute globules in another liquid by adding emulsifying agent.



CLASSIFICATION OF EMULSIONS:-

Emulsions can be classified into the following types:-

1. Oil in water (o/w) type of emulsion.
2. Water in oil (w/o) type of emulsion.
3. Microemulsions
4. Multiple/double emulsion.

▶ ADVANTAGES:-

- ▶ Mask the unpleasant taste.
- ▶ Sustained release medication.
- ▶ Inert and chemically non-reactive.
- ▶ Reasonably odourless & cost effective.

DISADVANTAGES:-

- Packing, handling & storage is difficult.
- Thermodynamically unstable & have short shelf life.
- Leads to creaming & cracking.
- Leads to phase inversion.

FORMULATION OF EMULSIONS:-

1. Selection of phases.
2. Phase volume ratio.
3. Choice of emulgent.
4. Antimicrobial agents.
5. Anti-oxidants.
6. Viscosifiers/consistency agents.
7. Colouring agents.
8. Sweetening agents.
9. Flavouring agents.
10. Emulsifying agents.

IDENTIFICATION TESTS:-

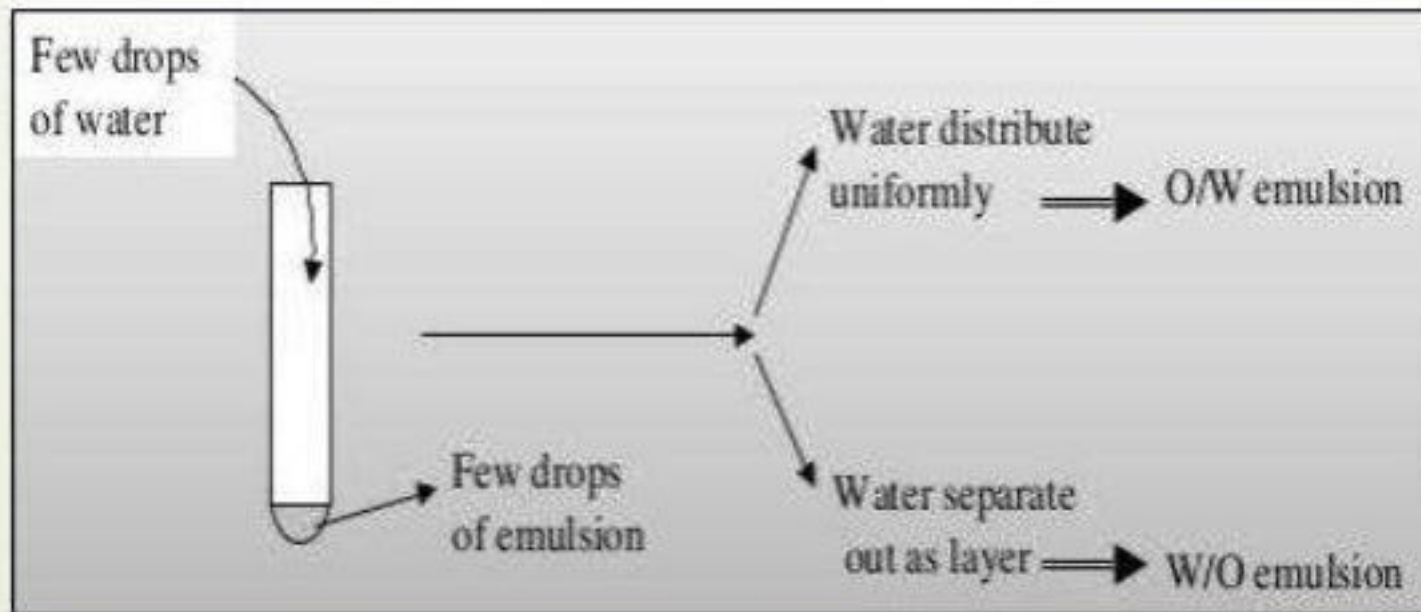
The type of emulsion can be determined by the following tests:-

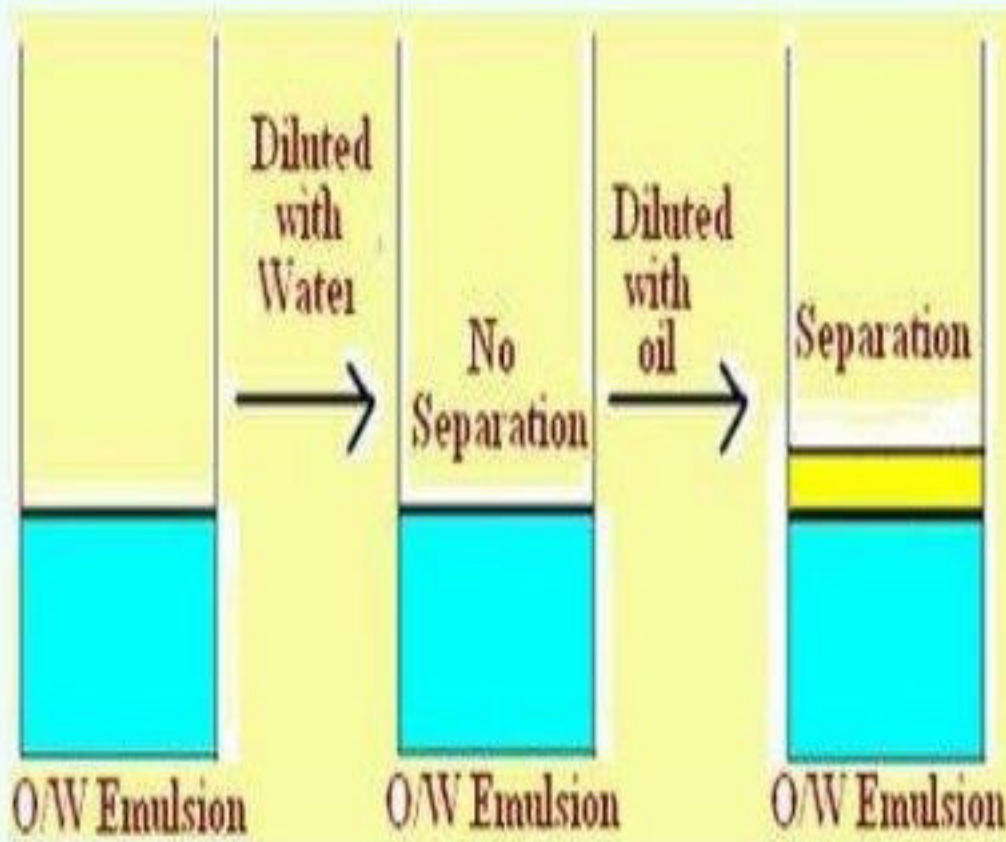
1. Dilution test.
2. Conductivity test.
3. Dye test.
4. Fluorescence test.
5. Cobalt chloride test (CoCl_2).

1. DILUTION TEST:-

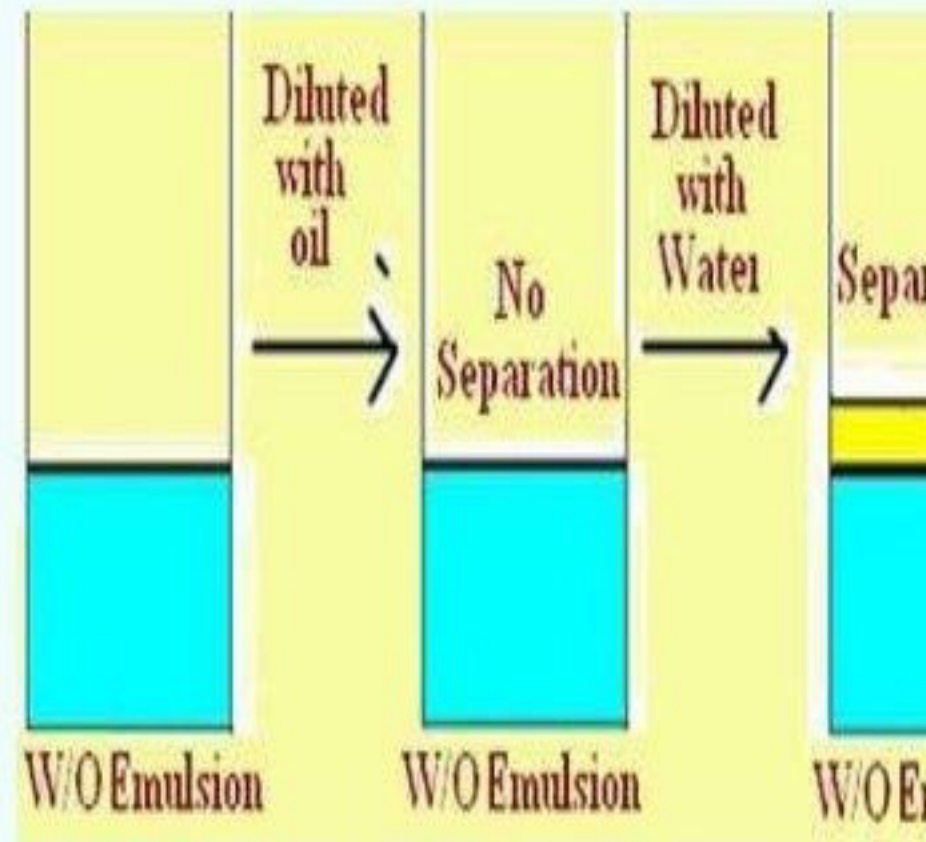
This test is based on the solubility of external phase of emulsion.

- o/w emulsion can be diluted with water.
- w/o emulsion can be diluted with oil.





Dilution Test for oil in water emulsion

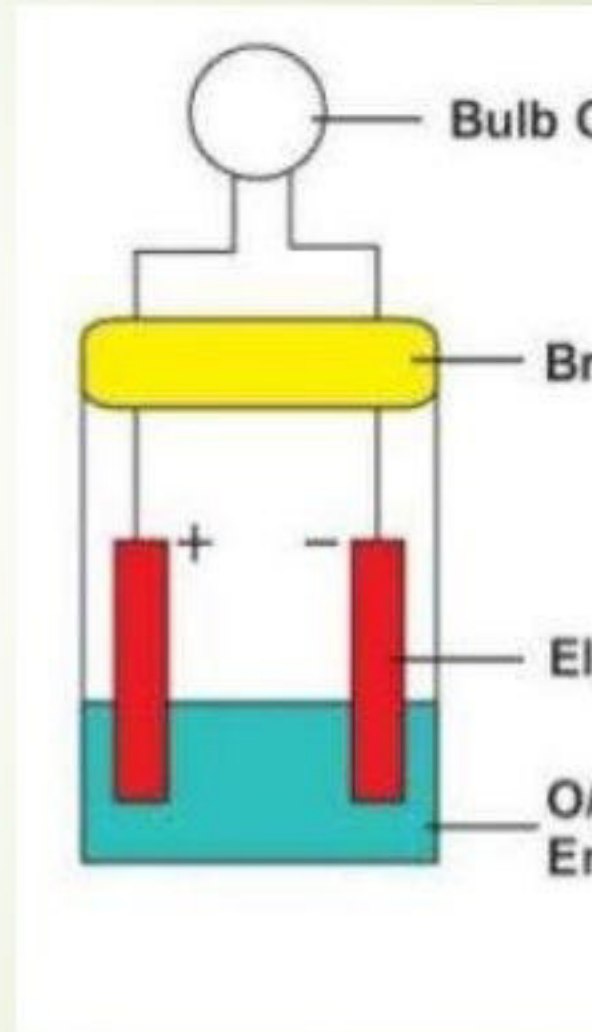


Dilution test for water in oil emulsion

2.CONDUCTIVITY TEST:-

The basic principle of this test is that water is a good conductor of electricity. Therefore in case of o/w emulsion this test will be +ve as water is the external phase.

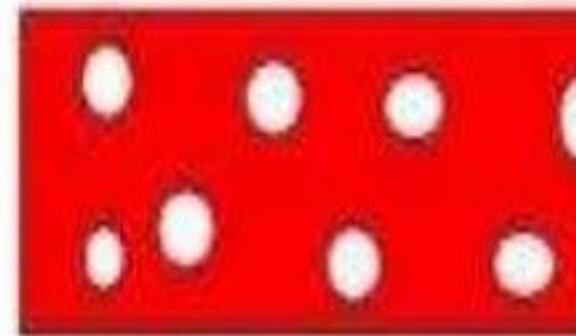
In this test,an assembly is used in which a pair of electrodes connected to an electric bulb is dipped into an emulsion.If the emulsion is o/w type,the electric bulb glows.



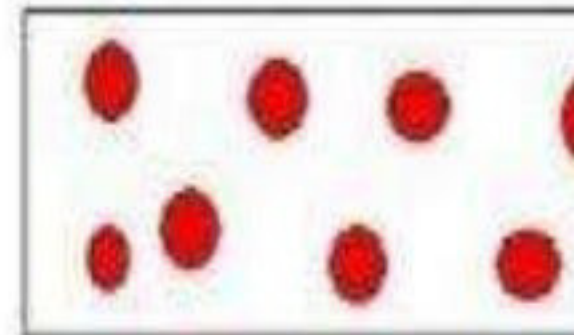
3.DYE TEST:-

When an emulsion is mixed with a water soluble dye such as amaranth and observed under the microscope.

- If the continuous phase appears red, then it means that the emulsion is o/w type as water is the external phase.
- If the scattered globules appear red and continuous phase is colourless, then it is w/o type.



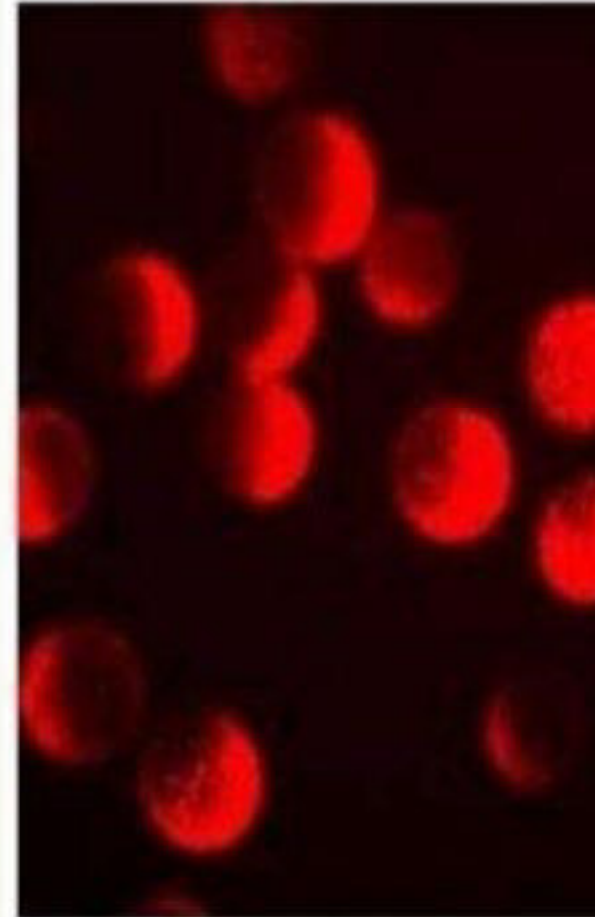
O/W Emulsion



W/O Emulsion

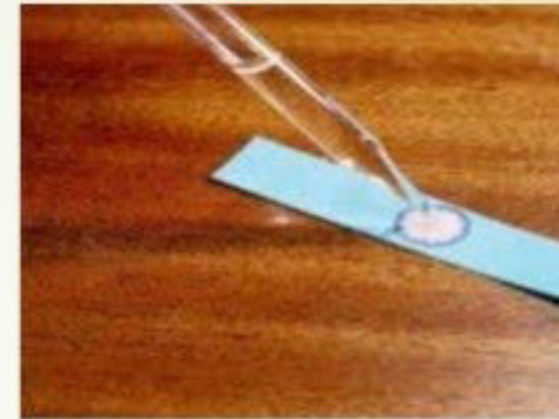
4.FLUORESCENCE TEST:-

Oil gives fluorescence under UV light, while water doesn't. Therefore, o/w emulsion shows spotty pattern when observed under UV, while w/o emulsion fluoresces.



5. COBALT CHLORIDE TEST:-

When a filter paper soaked in cobalt chloride solution is dipped into an emulsion and dried, it turns from blue to pink, indicating that the emulsion is o/w type.



Blue cobalt-chloride paper turning pink in the presence of water



PREPARATION OF EMULSIONS:-

The emulsions are prepared by two methods:-

1. Small scale method
 - a) Dry gum method
 - b) Wet gum method
 - c) Bottle method.
2. Large scale method.

EVALUATION OF EMULSIONS:-

1. Size distribution analysis.
2. Rate of phase separation.
3. Viscosity & rheological study.
4. Measurement of dielectric constant.
5. Conductivity measurement.
6. Influence of temperature.
7. Microwave radiation.
8. Microelectrophoretic measurement.

STABILITY OF EMULSIONS:-

The following three changes usually occurs during the storage of emulsion:-

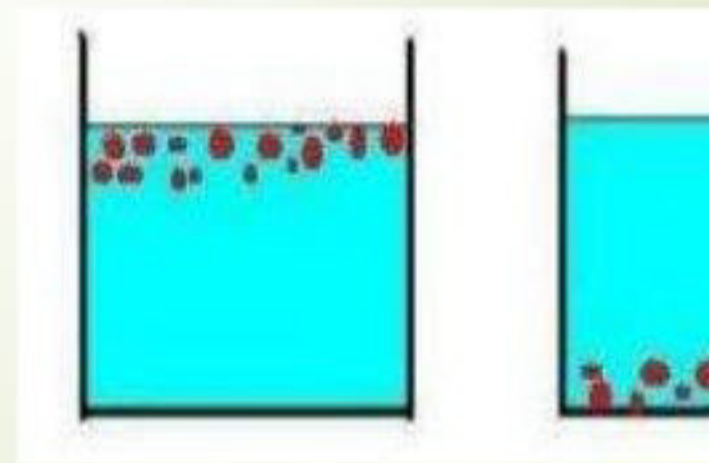
1. Creaming.
2. Cracking.
3. Phase inversion.

1.CREAMING:-

Creaming may be defined as the upward movement of dispersed globules to form a thick layer at the surface of emulsion.

The creaming depends on "Stokes law", the rate of creaming depends on the various factors.

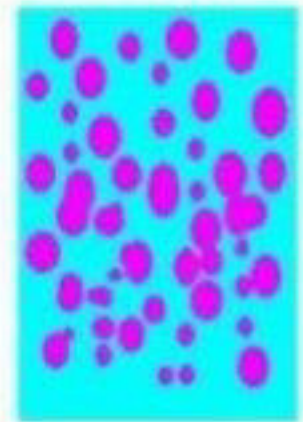
$$V = \frac{2r^2(d_1 - d_2)g}{9\eta}$$



2.CRACKING:-

Cracking means the separation of two layers of dispersed phase and continuous phase due to coalescence of dispersed phase globules. Cracking may be due to the following reasons:-

- By addition of emulsifying agent of opposite type.
- By decomposition of emulsifying agent.
- By addition of common solvent.
- By micro organisms.
- Changes in temperature.



Proper mixed

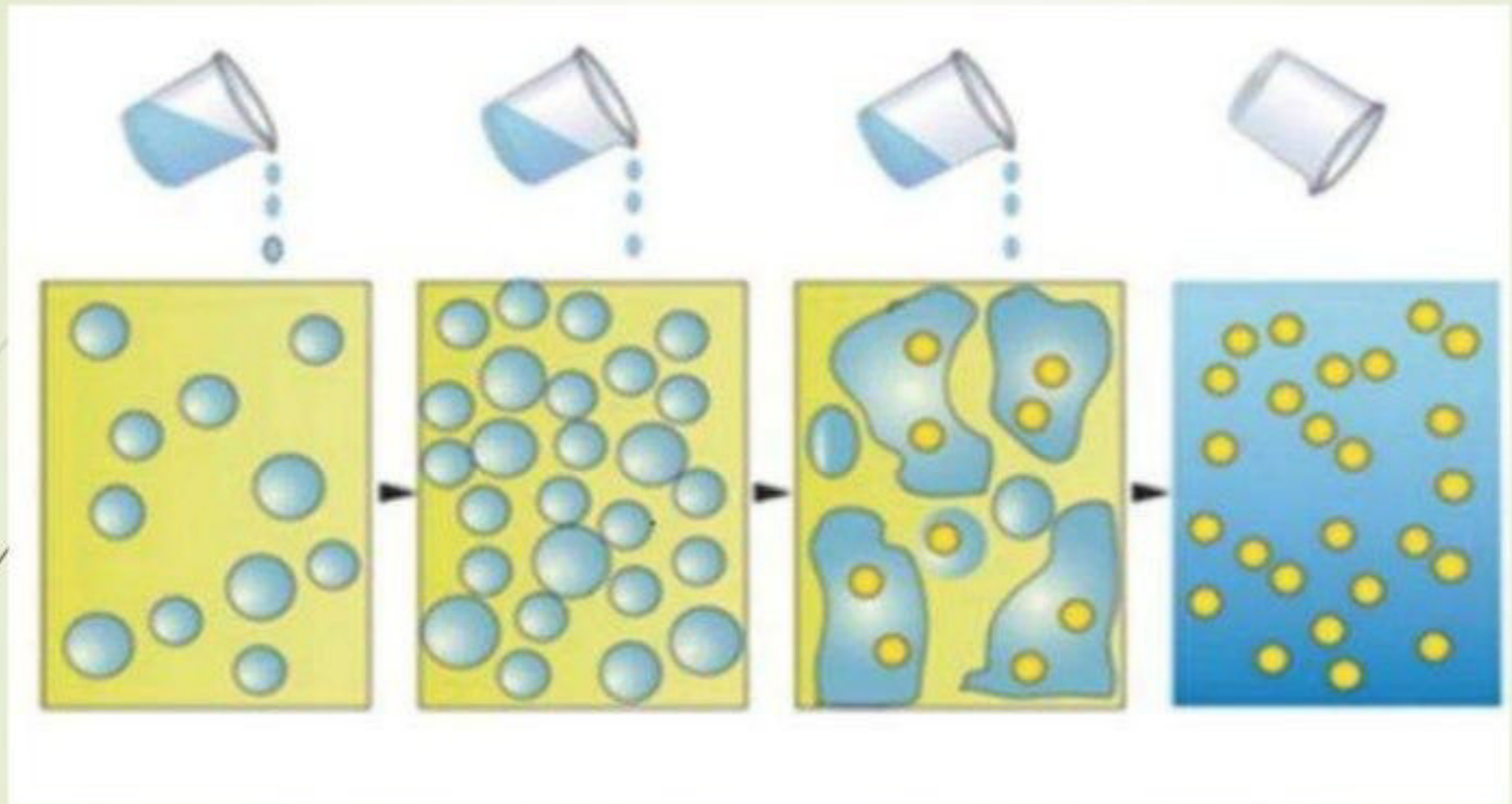


Phase

3.PHASE INVERSION:-

Phase inversion means change of one type of emulsion into the other type i.e., o/w emulsion changes into w/o type and vice versa.It may be due following reasons:-

- a)By the addition of an electrolyte.
- b)By changing the phase volume ratio.
- c)By temperature change.
- d)By changing the emulsifying agent.



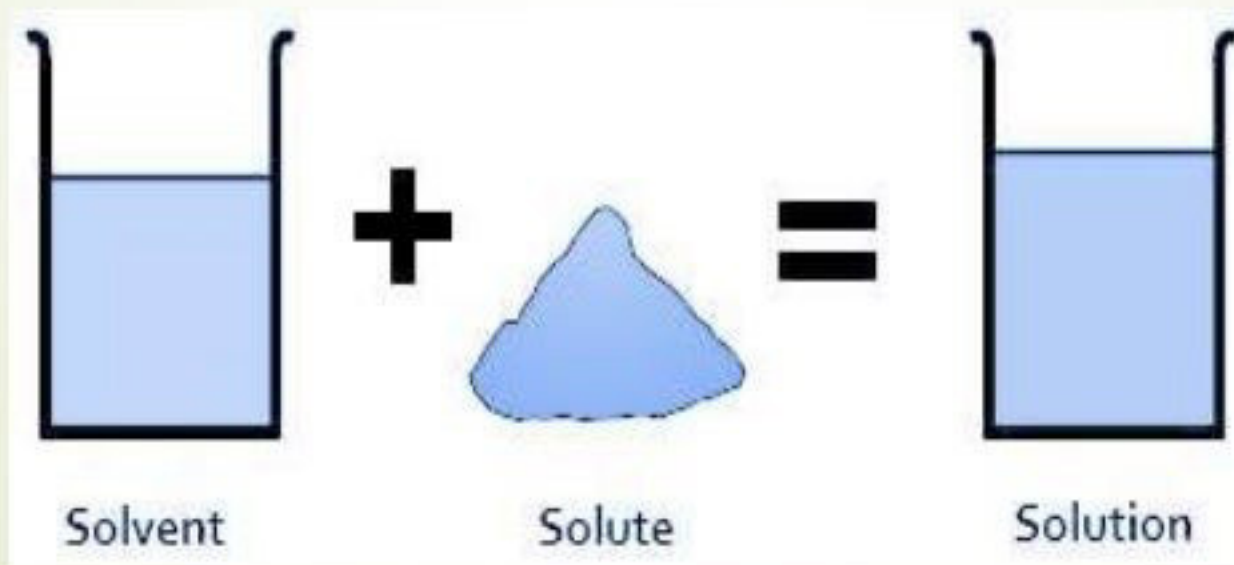
Phase inversion

SOLUTIONS



DEFINITION:-

In pharmaceutical terms, solutions are liquid preparations that contain one or more chemical substances dissolved in a suitable solvent or mixture of mutually miscible solvents.



▶ ADVANTAGES:-

1. Drug available immediately for absorption i.e., bioavailability of solutions is greater than that of oral solid dosage forms.
2. Flexible dosing.
3. Designed for oral route of administration.
4. No need to shake container.
5. Facilitates swallowing in difficult cases.

DISADVANTAGES:-

1. Drug stability reduced in solutions.
2. Bulky, difficult to transport and prone to container breakages.
3. Technical accuracy needed to measure dose on administration.
4. Measuring device is needed for administration.
5. Some drugs are poorly soluble.

CLASSIFICATION OF SOLUTIONS:-

- Oral solutions
- Oral syrups
- Oral elixirs
- Linctus
- Mouth washes/gargles

1. ORAL SOLUTIONS:-

- Oral solutions are administered to the GIT to provide absorption of the therapeutic agent.
- Oral solutions formulated over a broad p^H range due to the flexibility of GI environment.
- The usual p^H of oral solutions is about 7.0, unless there are issues regarding the solubility or stability of drugs.



2. ORAL SYRUPS:-

Syrups are highly concentrated aqueous solutions of sugar or a sugar substitute that contain a flavouring agent.

Eg:-Cherry syrup, orange syrup, raspberry syrup.



3. ORAL ELIXIRS:-

Elixir is a clear, hydroalcoholic solution formulated for oral use. The presence of alcohol in elixirs causes a problem in paediatric formulations and for adults who wish to avoid alcohol. It usually contains:-

- ▶ Purified water
- ▶ Alcohol
- ▶ Polyol cosolvents



4.LINCTUS:-

- A liquid oral preparation used for a demulcent, expectorant or sedative effect in treatment of cough.
- Linctuses are viscous preparations that contain the therapeutic agent dissolved in a vehicle composed of a high percentage of sucrose or other sweetening agents.
- Primarily employed for the treatment of cough, due to their soothing actions on the inflamed mucous membranes.



5. MOUTH WASHES/GARGLES:-

- These are designed for the treatment of infections and inflammation of the oral cavity.
- Formulations designed for this purpose employ water as the vehicle, although a cosolvent (alcohol) may be employed to solubilize the active agent.
- They include preservatives, colouring and flavouring agents and sweetening agents.



STABILITY OF SOLUTIONS:-

Both the chemical and physical stability of solution in their container are important. A solution must retain its initial clarity, colour, odour, taste and viscosity over its allocated shelf life.

Major signs of instability are:-

1. Colour change
2. Precipitation
3. Microbial growth
4. Chemical gas formation.

Thank
you
Love

TABLETS

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- 
- Introduction
 - General Properties
 - Advantages
 - Disadvantages
 - Types of Tablets
 - Tablet Additives

Formulation Development

- Preformulation of drugs & additives
- Introduction to tablet Additives
- Need of Granulation
- Mechanisms
- Manufacturing Processes and Equipments for granulation
- Advance Granulation Techniques
- Characterization and Evaluation

All about Tablets.

All about Tablet Coating

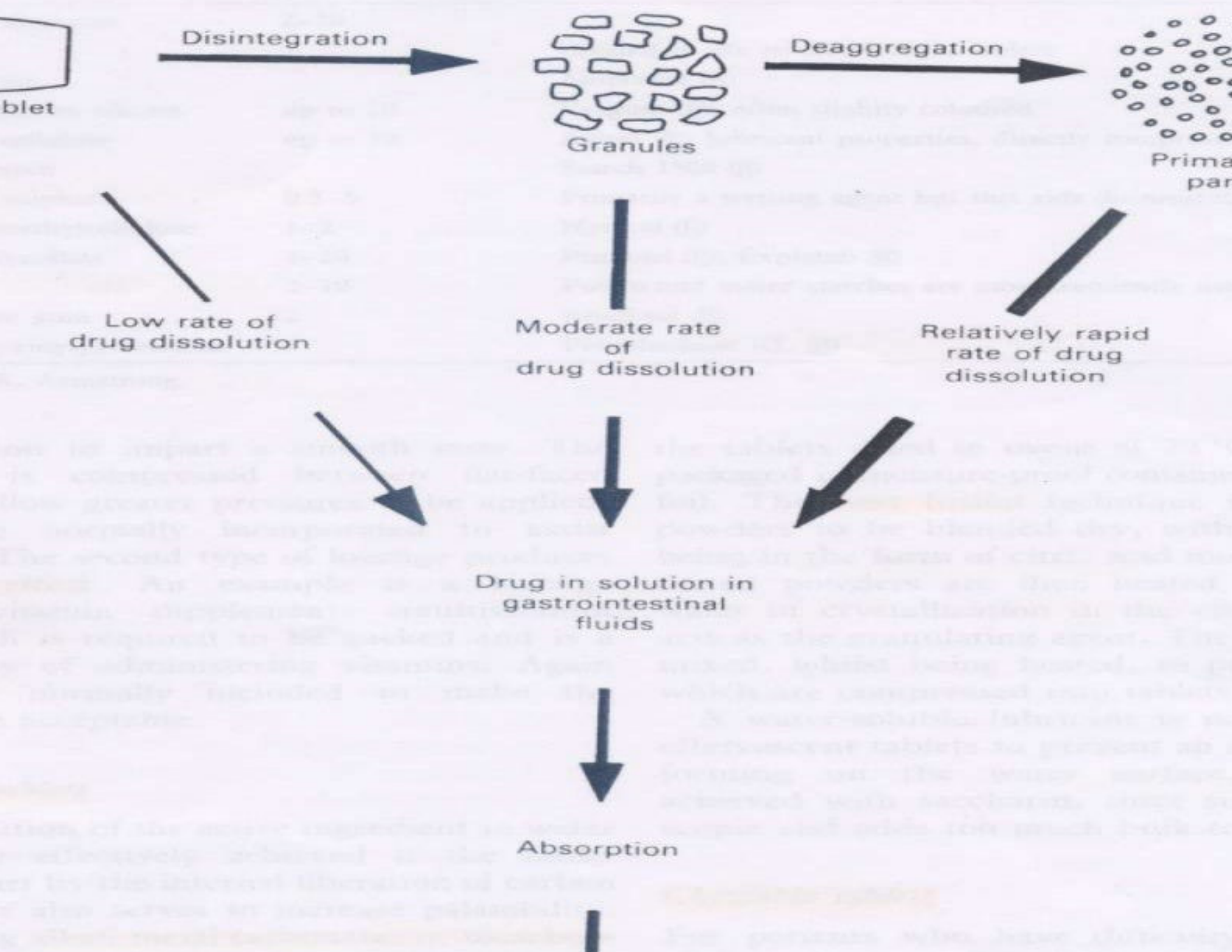
Introduction

Definition:

Tablet is defined as a compressed unit solid dosage form containing medicaments with or without excipients.

According to the Indian Pharmacopoeia, pharmaceutical tablets are solid, flat or biconvex dishes, unit dosage form, prepared by compressing a drug or a mixture of drugs, with or without diluents.

They vary in shape and differ greatly in size and weight, depending on amount of medicinal substances and the intended mode of administration.



➤ *Advantages*

greatest dose precision and the least content variability.

Cost is lowest of all oral dosage forms.

Lighter and compact.

Easiest and cheapest to package and strip.

Easy to swallow with least tendency for hang-up.

tinued...

Sustained release product is possible by various techniques.

Objectionable odour and bitter taste can be masked by various techniques.

Suitable for large scale production.

Greatest chemical and microbial stability over all oral dosage forms.

Product identification is easy and rapid requiring no additional

when employing an embossed and/or monogrammed punch face.

Disadvantages

Difficult to swallow in case of children and unconscious patients.

Some drugs resist compression into dense compacts, owing to amorphous, low density character.

Drugs with poor wetting, slow dissolution properties, may be difficult to formulate or manufacture as a tablet that will still provide adequate or full bioavailability.

For taste testing drugs, drugs with an objectionable odor or drugs that are sensitive to moisture may require encapsulation or coating. In such cases, capsule manufacturing is often the most convenient and lowest cost.

General properties of Tablet dosage forms

Tablet should have elegant product identity while free of defects like chips, cracks, discoloration, and contamination.

Tablet should have sufficient strength to withstand mechanical shock during its production, storage, shipping and dispensing.

Tablet should have the chemical and physical stability to maintain its physical attributes over its shelf life.

Tablet must be able to release the medicinal agents in a predictable and reproducible manner.

Tablet should have a chemical stability over time so as not to follow alteration of the medicinal agent.

Tablets ingested orally:

- Compressed tablet, e.g. Paracetamol tablet
- Multiple compressed tablet
- Delayed release tablet, e.g. Enteric coated Bisacodyl tablet
- Sugar coated tablet, e.g. Multivitamin tablet
- Film coated tablet, e.g. Metronidazole tablet
- Chewable tablet, e.g. Antacid tablet

Tablets used in oral cavity:

- Buccal tablet, e.g. Vitamin-c tablet
- Sublingual tablet, e.g. Vicks Menthol tablet
- Troches or lozenges

Tablets administered by other route:

- Implantation tablet
- Suppositories or Inserts, e.g. Clotrimazole tablet

Tablets used to prepare solution:

- Effervescent tablet, e.g. Dispirin tablet (Aspirin)
- Dispensing tablet, e.g. Enzyme tablet (Digiplex)
- Hypodermic tablet
- Tablet triturates e.g. Enzyme tablet (Digiplex)

Tablet Additives

In addition to active ingredients, tablet contains a number of inert materials known as additives or excipients. Different excipients are:

Filament

Binder and adhesive

Disintegrants

Lubricants and glidants

Coloring agents

Flavoring agents

Sweetening agents

excipients are *fillers* used to make *required bulk* of the tablet when the drug dosage is inadequate to produce the bulk.

Secondary reason is *to provide better tablet properties* such as *improve cohesiveness* to permit use of *direct compression* manufacturing or to *promote flow*.

Excipients should have following properties:

They must be non toxic

They must be commercially available in acceptable grade

Their cost must be low

They must be physiologically inert

They must be physically & chemically stable by themselves & with the drug

They must be free from all microbial contamination.

They do not alter the **bioavailability of drug**.

Lactose-anhydrous and spray dried lactose

Directly compressed starch-Sta Rx 1500

Hydrolyzed starch-Emdex and Celutab

Microcrystalline cellulose-Avicel (PH 101 and PH 102)

Dibasic calcium phosphate dehydrate

Calcium sulphate dihydrate

Mannitol

Sorbitol

Sucrose- Sugartab, DiPac, Nutab

Dextrose

Binders and Adhesives:

materials are added either dry or in wet- form to form granules or cohesive compacts for directly compressed tablet.

Example: Acacia, tragacanth- Solution for 10-25% Conc.

Cellulose derivatives- Methyl cellulose, HPC, HPMC

Gelatin- 10-20% solution

Glucose- 50% solution

Polyvinylpyrrolidone (PVP)- 2% conc.

Starch paste-10-20% solution

Sodium alginate

Sorbitol

Disintegrants:

used in a tablet formulation to facilitate its breaking or disintegration upon contact with water in the GIT.

Example: Starch- 5-20% of tablet weight.

Starch derivative – Primogel and Explotab (1-8%)

Clays- Veegum HV, bentonite 10% level in colored tablet only

Cellulose

Cellulose derivatives- Ac-Di-Sol (sodium carboxy methyl cellulose)

Alginate

PVP (Polyvinylpyrrolidone), cross-linked

swell up to ten fold within 30 seconds when contact water.

Example: Crosscarmellose- cross-linked cellulose,
Crosspovidone- cross-linked povidone (polymer),
Sodium starch glycolate- cross-linked starch.

cross-linked products swell upto 10 fold with in 30 seconds when in contact with water.

portion of disintegrant is added before granulation and a portion before compression, which serve as **glidants or lubricant**.

generation of carbon dioxide in effervescent tablets is also one way of disintegration

Lubricant and Glidants:

Lubricants are intended to prevent adhesion of the tablet material

to the surface of dies and punches,

Glidants are intended to promote flow of granules or powder material by reducing the friction between the particles.

Example:

Lubricants-

Stearic acid, Stearic acid salt - Stearic acid, Magnesium stearate, Talc, Polyethylene glycols), Surfactants

Glidants-

Starch – 5-10% conc., Talc-5% conc., Silica derivative - Colloidal silicas such as

The use of colors and dyes in a tablet has three purposes:

Masking of off color drugs

Product Identification

Production of more elegant product

All coloring agents must be approved and certified by FDA. Two forms of color are used in tablet preparation – FD &C and D & C dyes. These dyes are applied as suspensions, granulating agent or Lake form of these dyes. Lakes are dyes absorbed on hydrophilic pigments and employed as dry powder coloring.

Example: FD & C yellow 6-sunset yellow

FD & C yellow 5- Tartrazine

FD & C green 3- Fast Green

FD & C blue 1- Brilliant Blue

FD & C blue 2 - Indigo carmine

Flavoring agents:

Chewable tablet- flavor oil are used

Sweetening agents:

Chewable tablets: Sugar, mannitol.

Saccharine (artificial): 500 times sweeter than sucrose

Disadvantage: Bitter aftertaste and carcinogenic

Aspartame (artificial)

Disadvantage: Lack of stability in presence of moisture.

Granulation

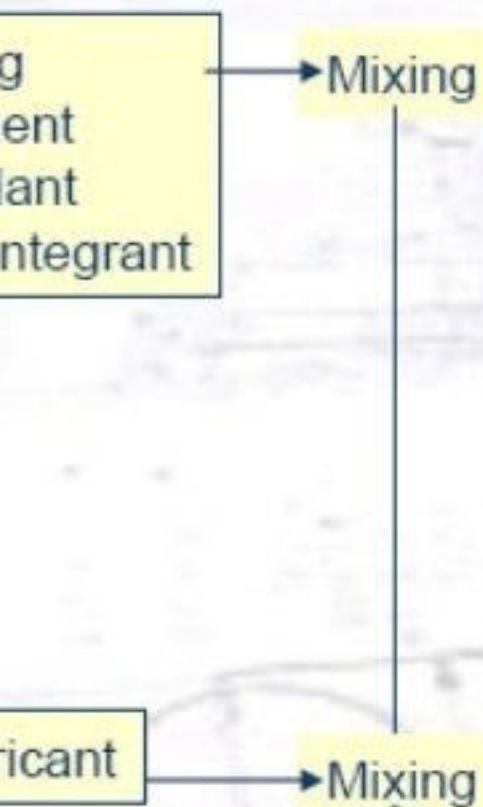
Need

- To prevent segregation of the constituents of the powder mix
- To improve the flow properties of the mix
- To improve the compaction characteristics of the mix
- Other reasons: Toxic, Slightly hygroscopic, denser.

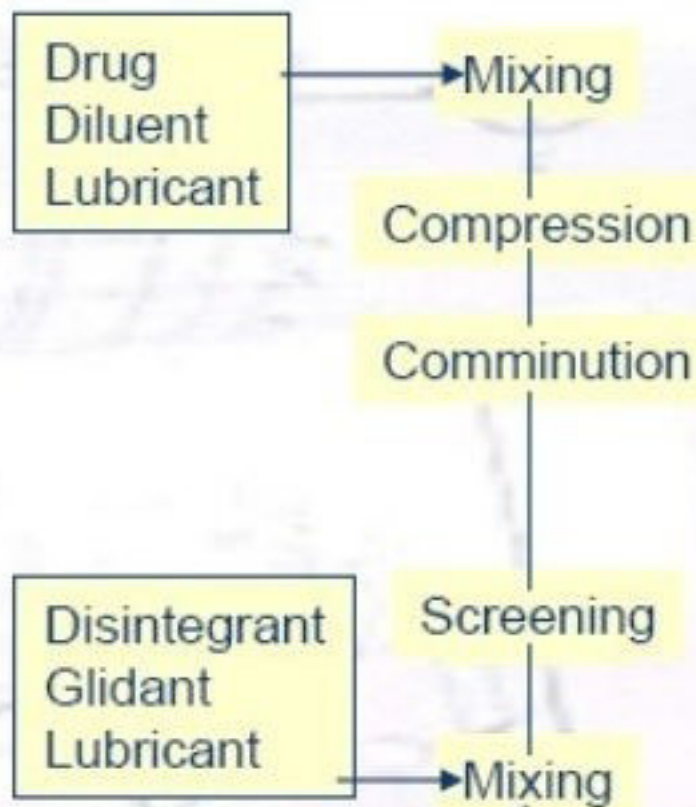
Methods

- Dry granulation
- Wet granulation

Direct Compression



Dry Granulation



Wet Granulation



Fill Die, Compress Tablet, Eject Tablet

Mechanisms of Granulation

There are Five Particle Bonding Mechanisms,

Adhesion and cohesion forces in the immobile liquid films

Interfacial forces in mobile liquid films within the granules

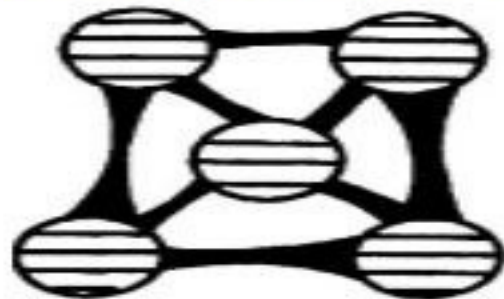
Formation of solid bridges after solvent evaporation

Attractive forces between solid particles

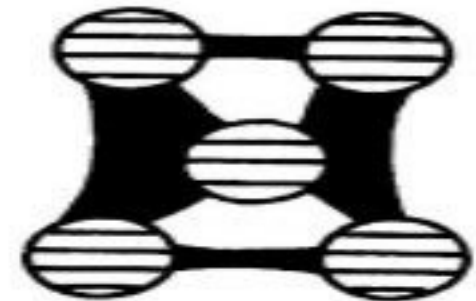
Mechanical interlocking

➤ *Adhesion and cohesion forces in immobile liquid films* between individual primary powder particles.

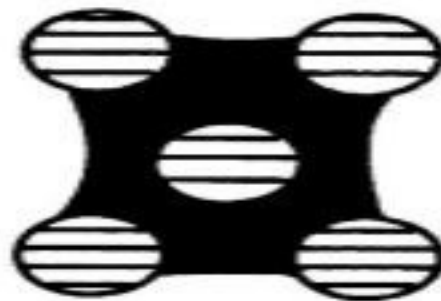
➤ *Interfacial forces in mobile liquid films*



Pendular



Funicular



Capillary



Droplet

➤ *Solid bridges*

Partial melting, Binder hardening, crystallization of dissolved sub.

➤ *Attractive forces* between *solid particles*

Weighing

Mixing

Granulation

Screening

Drying

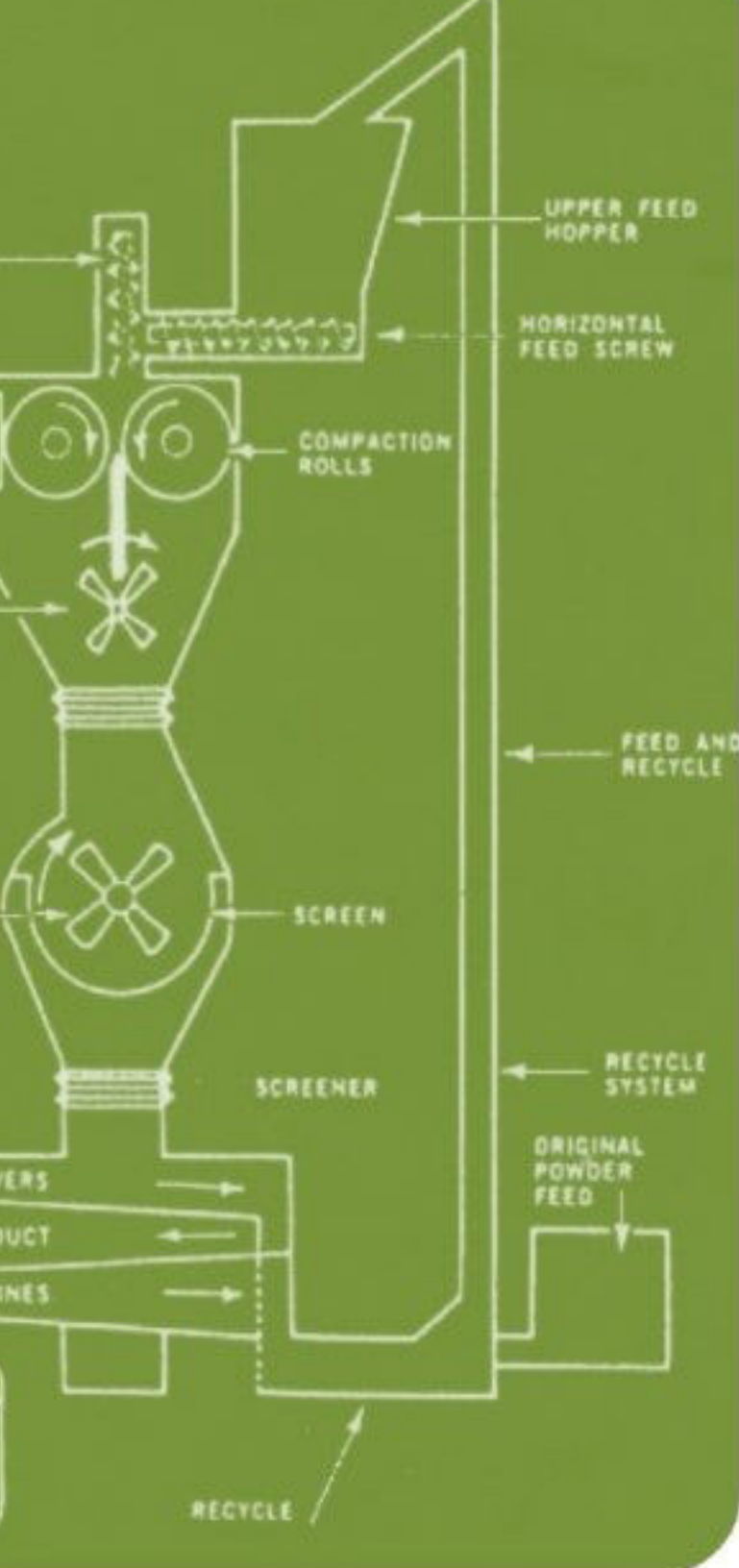
Screening

Lubrication

Compression

(Granulators)

- Dry granulator
- Wet granulator



□ Dry granulators:

- Sluggers
- Roller Compactors

Is used when.....

- Effective dose of drug is too high for direct compression
- Drug is sensitive to heat or moisture or both.

Wet granulators

Shear mixer granulator

High speed granulator

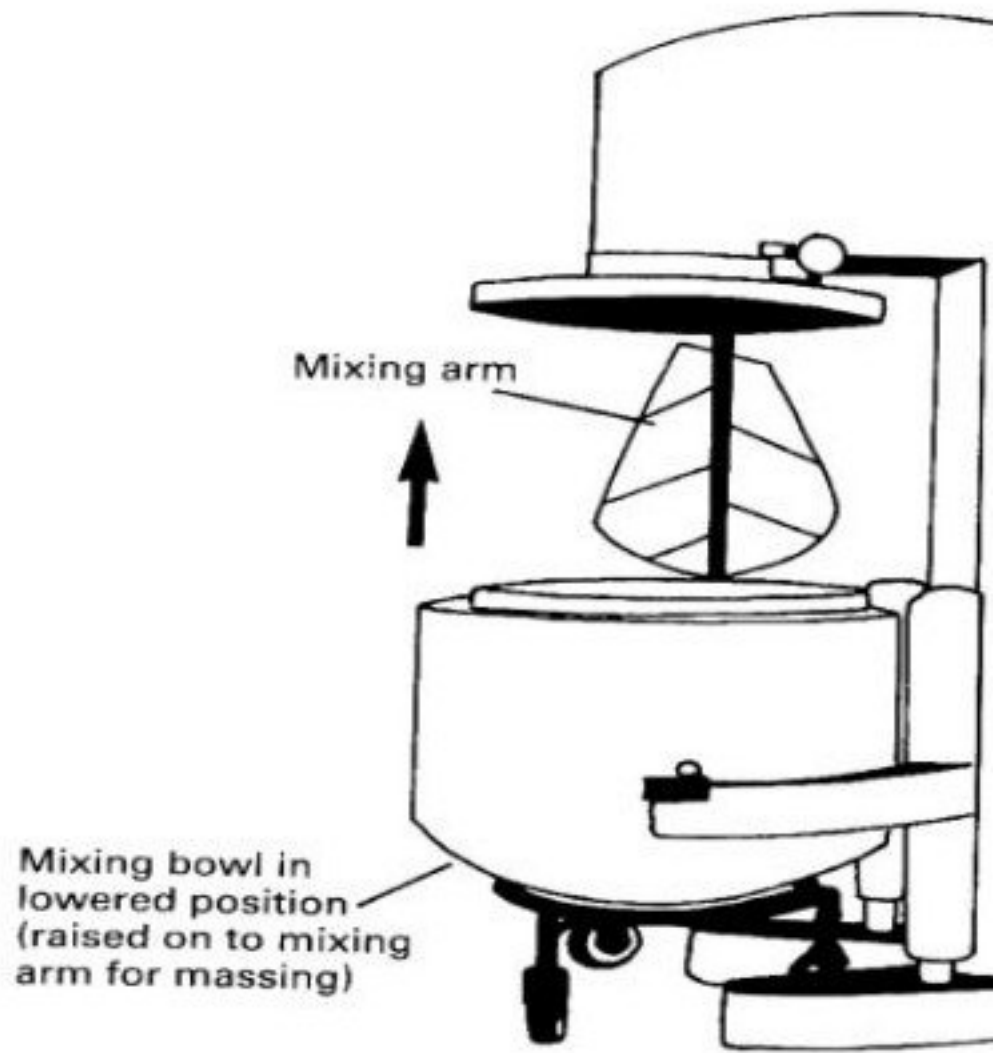
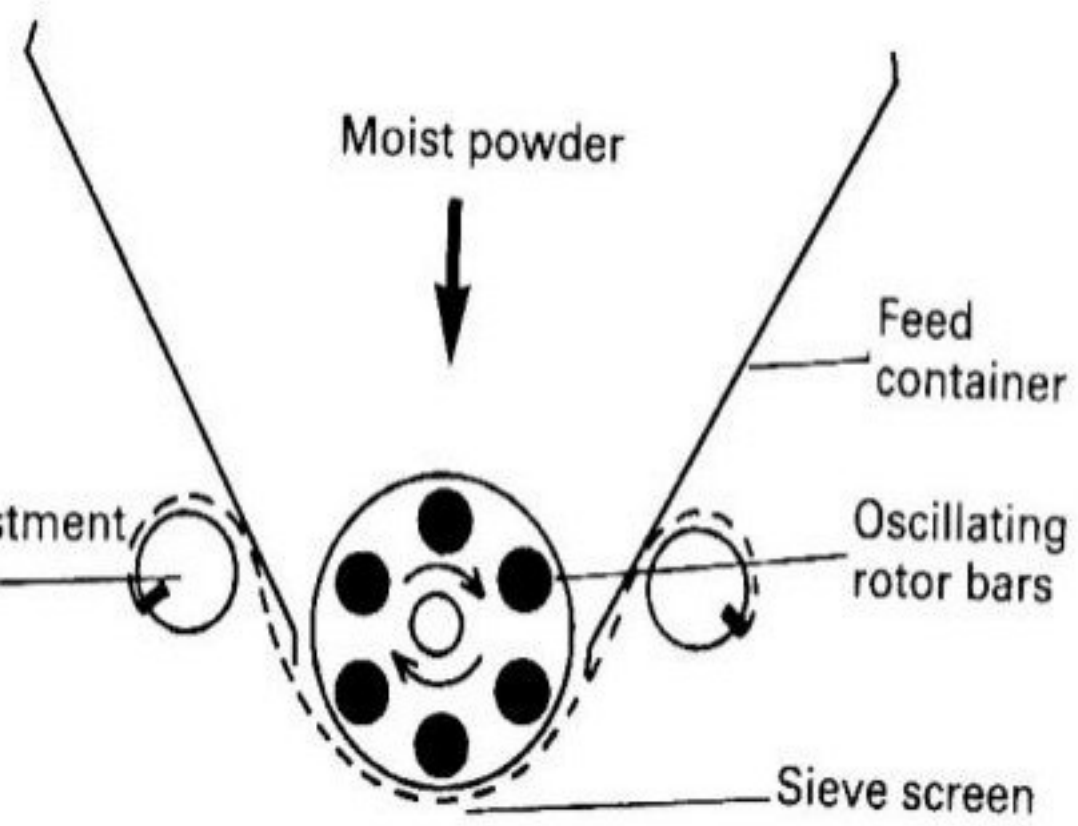
Fluidized bed granulator

Spray driers

High Shear Granulator	Littleford Lodige Mixer/Granulator Littleford MGT Granulator Diosna Granulator Gral Mixer/Granulator
Granulator with Drying Facility	Fluidized Bed Granulator Day Nauta Mixer Processor Double cone/Twin Shell Processor Topo Granulator
Special Granulator	Roto Granulator Marumerizer

Wet granulation equipment

- shear granulator



OP Of Shear Mixture Granulator

Mixed powder are fed in to the bowl

Granulating liquid is added

The moist mass has then transferred to a granulator such as oscillating granulator

Disadvantage

long duration

large number of equipment are needed

high material loss

Advantage

not very sensitive to the material

end point can be determined by inspection

High speed granulator

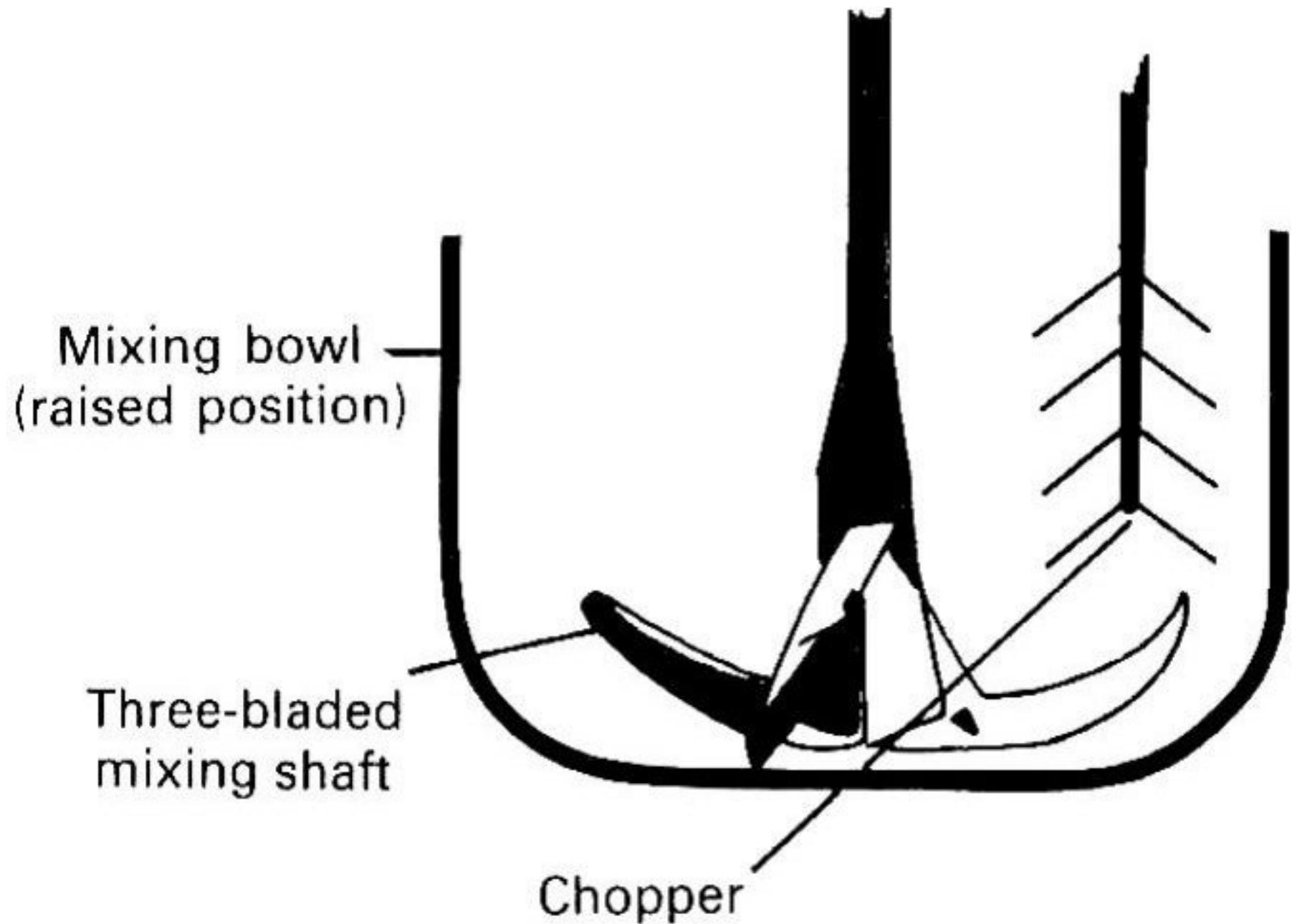
Widely used in pharmaceutical

SS mixing bowl containing a three blade main impeller, revolves in horizontal plane, and a three blade auxiliary chopper –revolves vertical or horizontal plane

Unmixed powder –in the bowl mixed for few minute with rotating impeller

Granulation

High speed granulator



Rapid Mixer Granulator (RMG)

gives more normal PSD

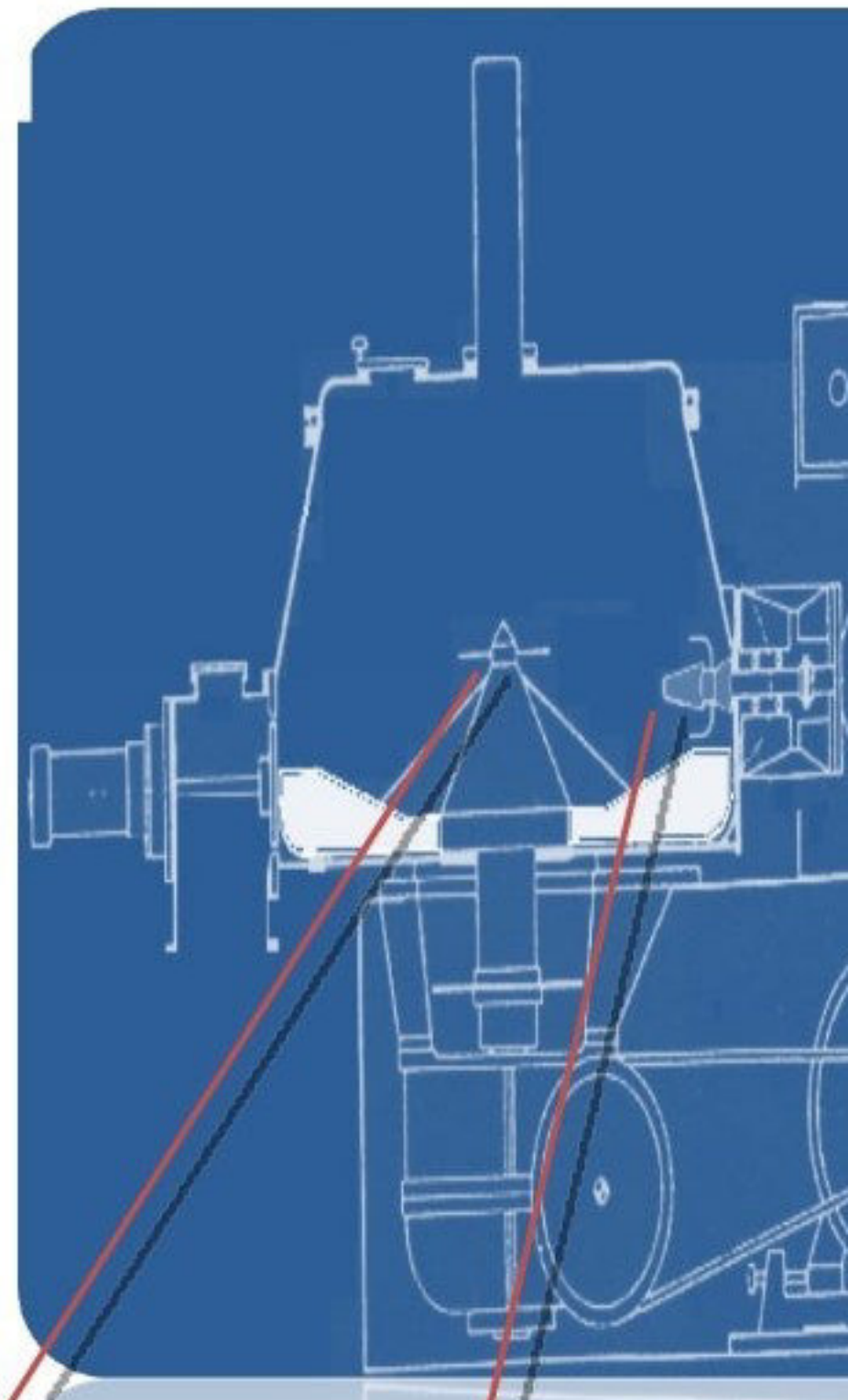
with lesser fines.

Typical Time Sequence

Mixing – 2 minutes

Granulation – 8 minutes

Discharge – 1 minutes





Advantage

Mixing, Massing, Granulation in a single equipment
within few minutes

Disadvantage

End point monitor needed

signs of FB granulators

Top spray

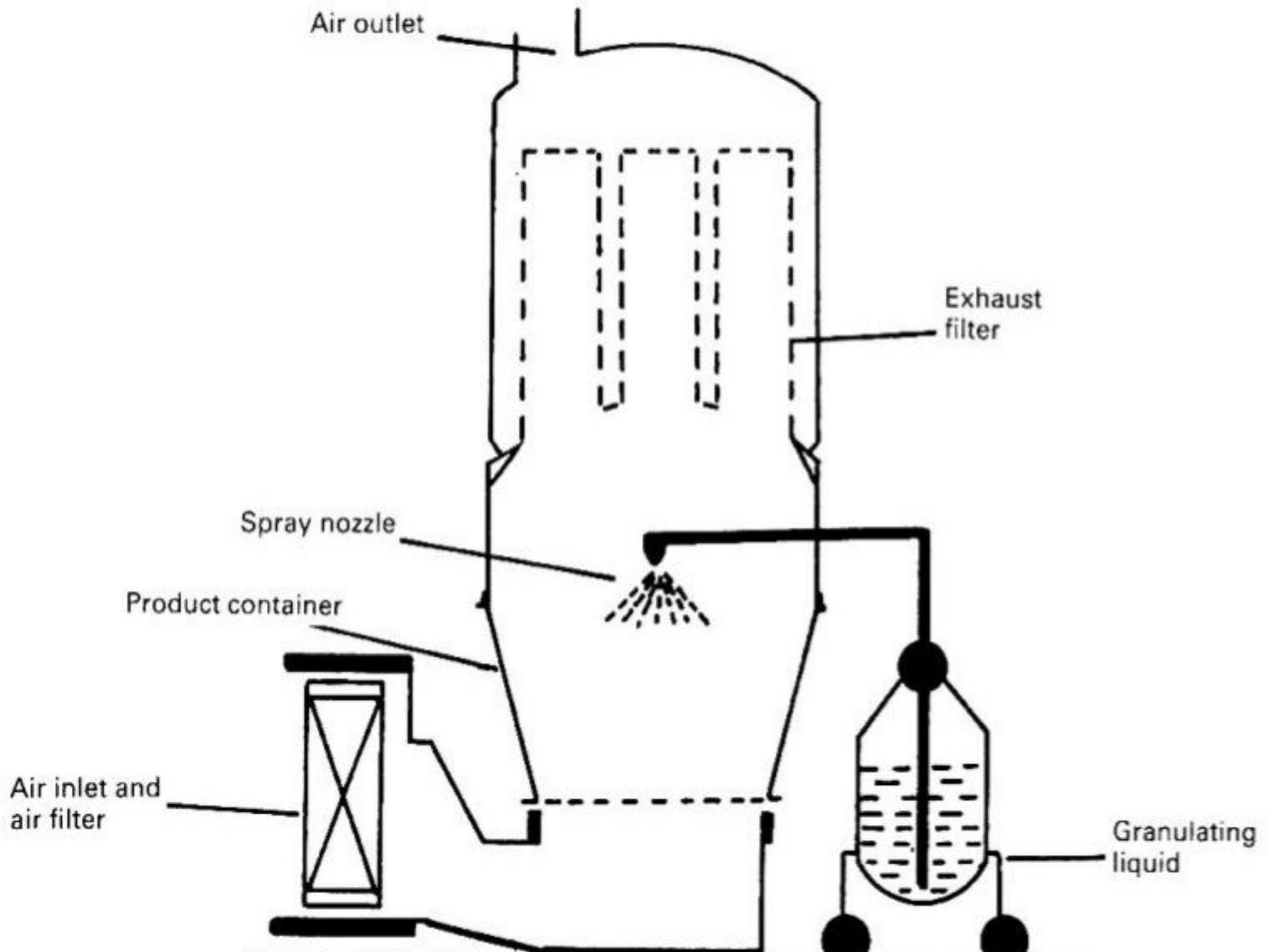
Bottom spray

Rotating disc

granulator



Fluidized Bed Granulator



Fluidized Bed Granulator

Advantage

- One unit so saving labour cost, transfer losses and time
- 2-6 time greater heat transfer than tray dryer
- Uniform drying....prevent mottling.
- Process can be automated once parameters optimized

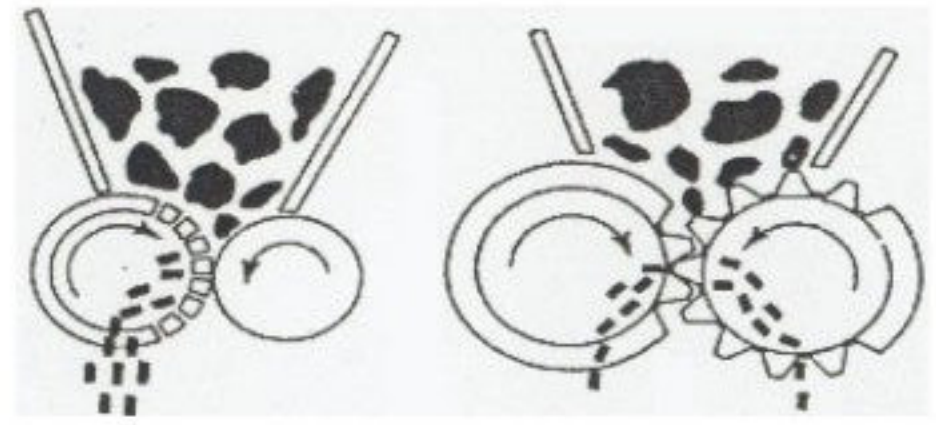
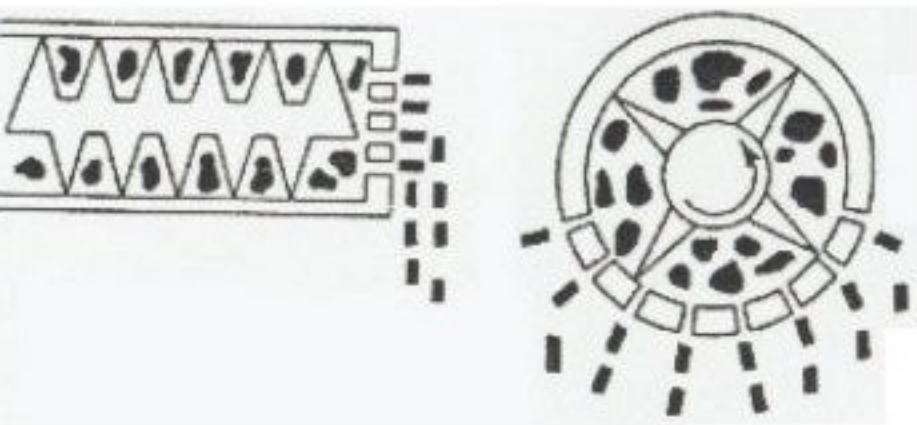
Disadvantage

- Expensive
- Multiple process variable
- Filter clogging, demixing, electrostatic charge, solvent explosion

Fluidized Bed Granulator (Industrial Equipment)



Net mass containing *drug*, *diluents* and *binder* is passed *through extruder* to get *rod* shaped segments.



Screw-feed Extruder

Cylinder

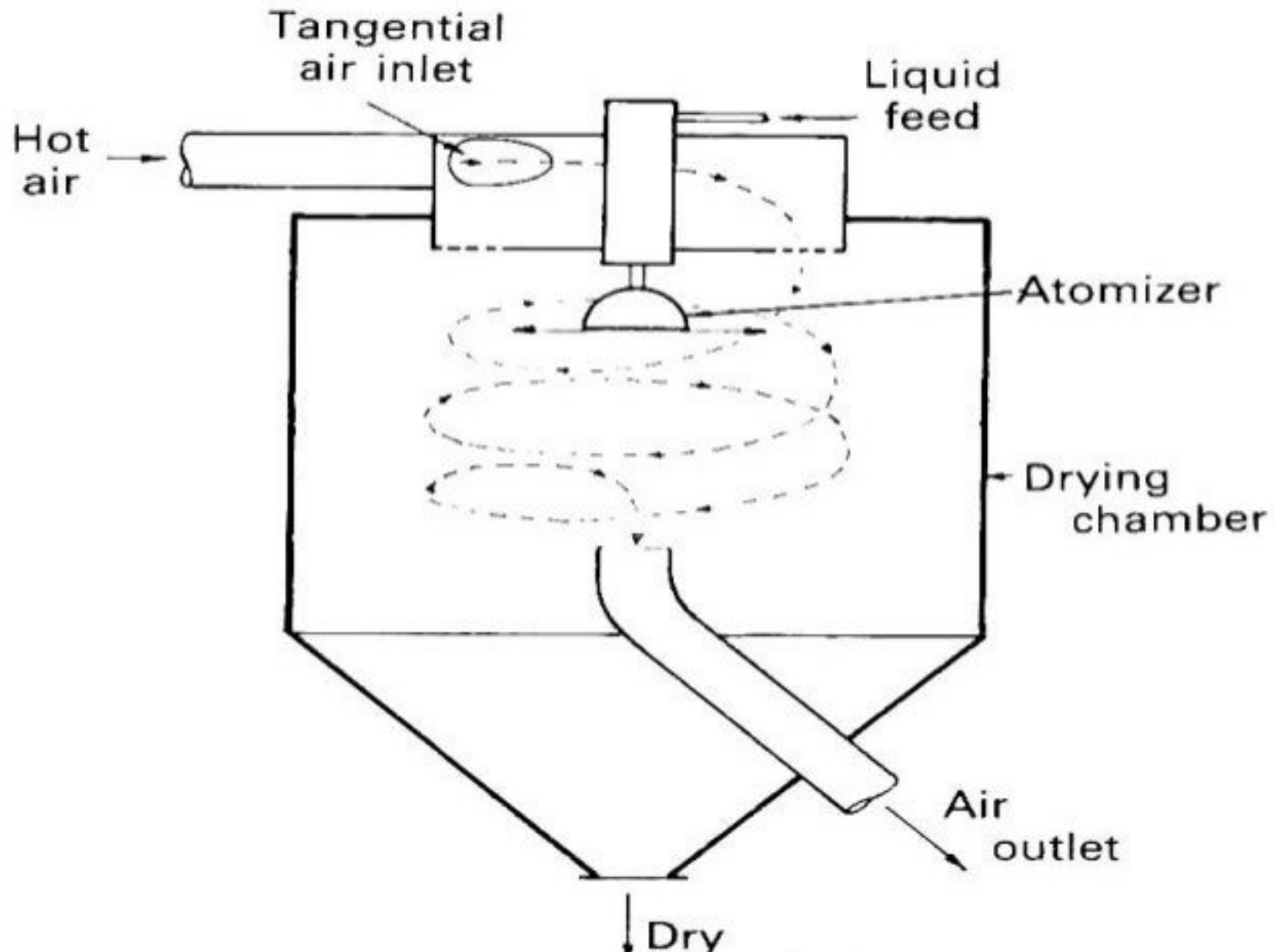
Segments are placed in GRANULATOR where they are shaped here by centrifugal and frictional forces produced by rotating blades and form granules

Advantage

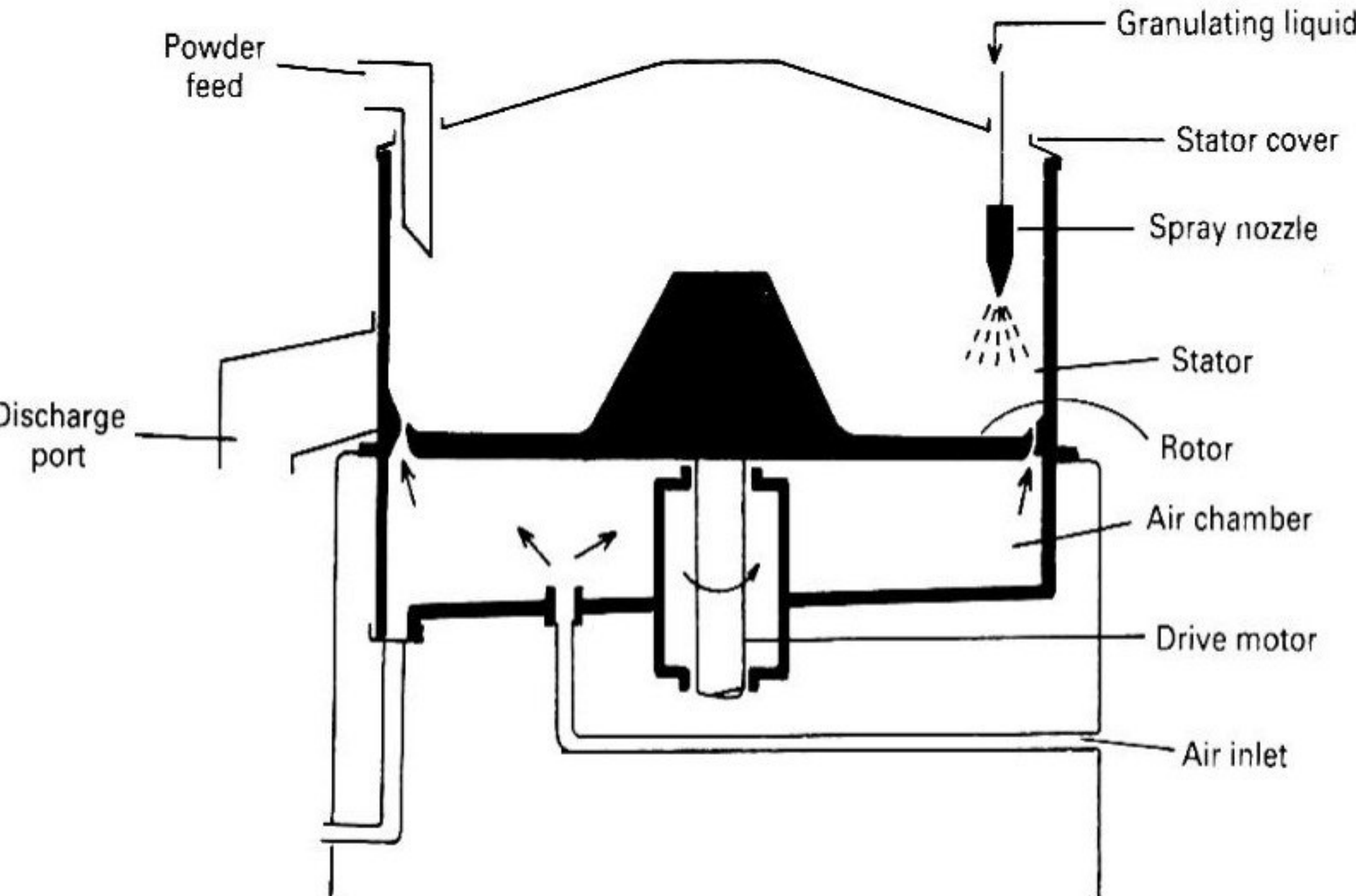
Granules with regular size, shape with lower friability,

Other More Specialized Granulators

- **Spray Driers**



■ Pelletizers



A melt granulation process has been investigated which efficiently agglomerates pharmaceutical powders for use in both immediate- and sustained-release solid dosage forms.

The process utilizes materials that are effective as granulating fluids when they are in the molten state.

Cooling of the agglomerated powders and the resultant solidification of the molten materials completes the granulation process.

Both the molten agglomeration and cooling solidification were accomplished in a high shear Collette Gral mixer equipped with a jacketed bowl.

Hence, the melt granulation process replaces the conventional granulation and drying operations which use water or alcohol

Need of other component

Large dose- not suitable

Small dose-impractical

Moderate dose- suitable

Directly compressible vehicles

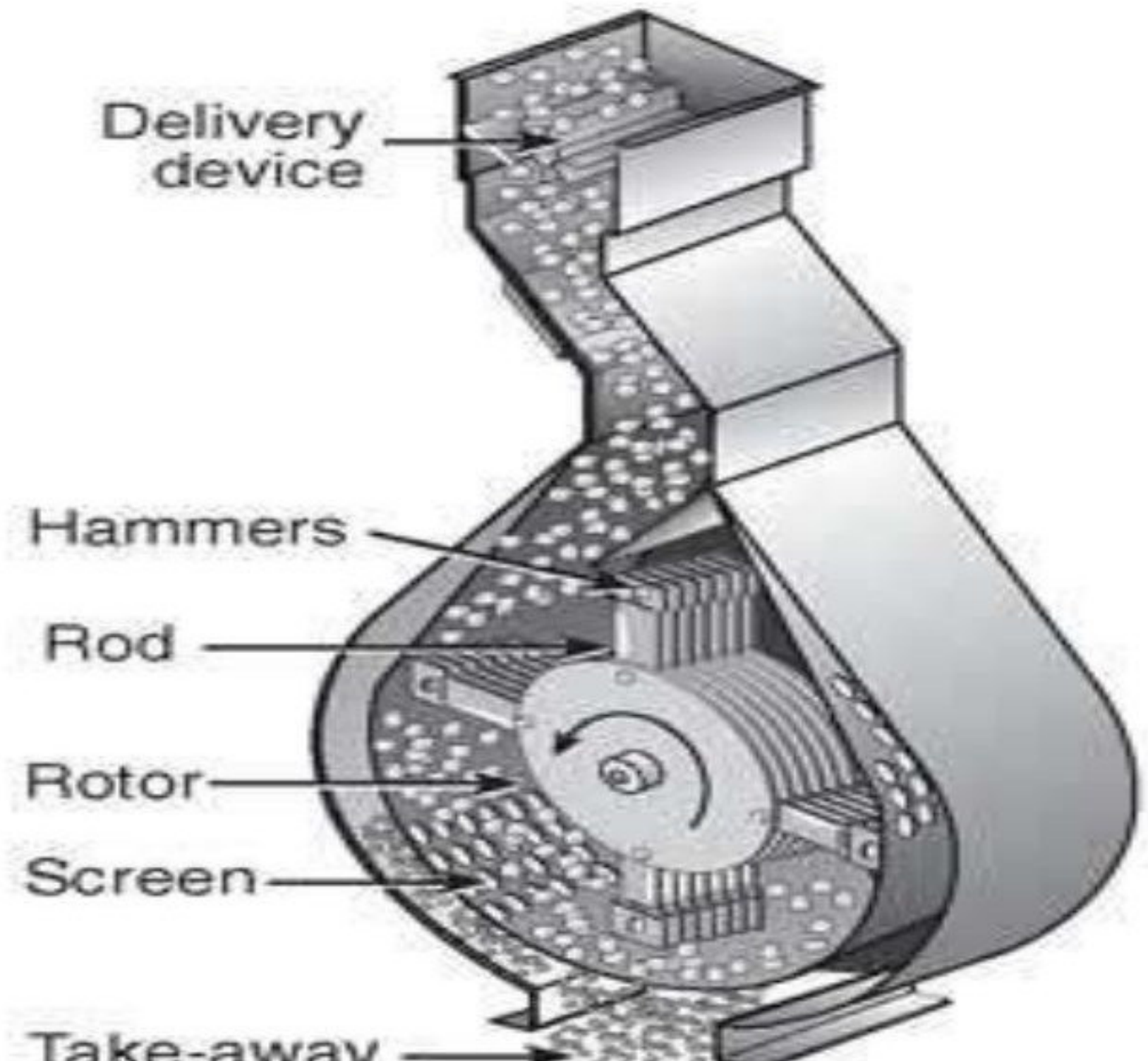
Limitations

- 1) Stratification-poor content uniformity
- 2) Large dose drugs [30%]
- 3) Interaction
- 4) Static charge

Equipment and procedures used

- 1) Screening/Milling
 - * Three Parts
 - * Principle
 - * Operation
 - * Types of Mills

Hammer Mill



Ball Mill



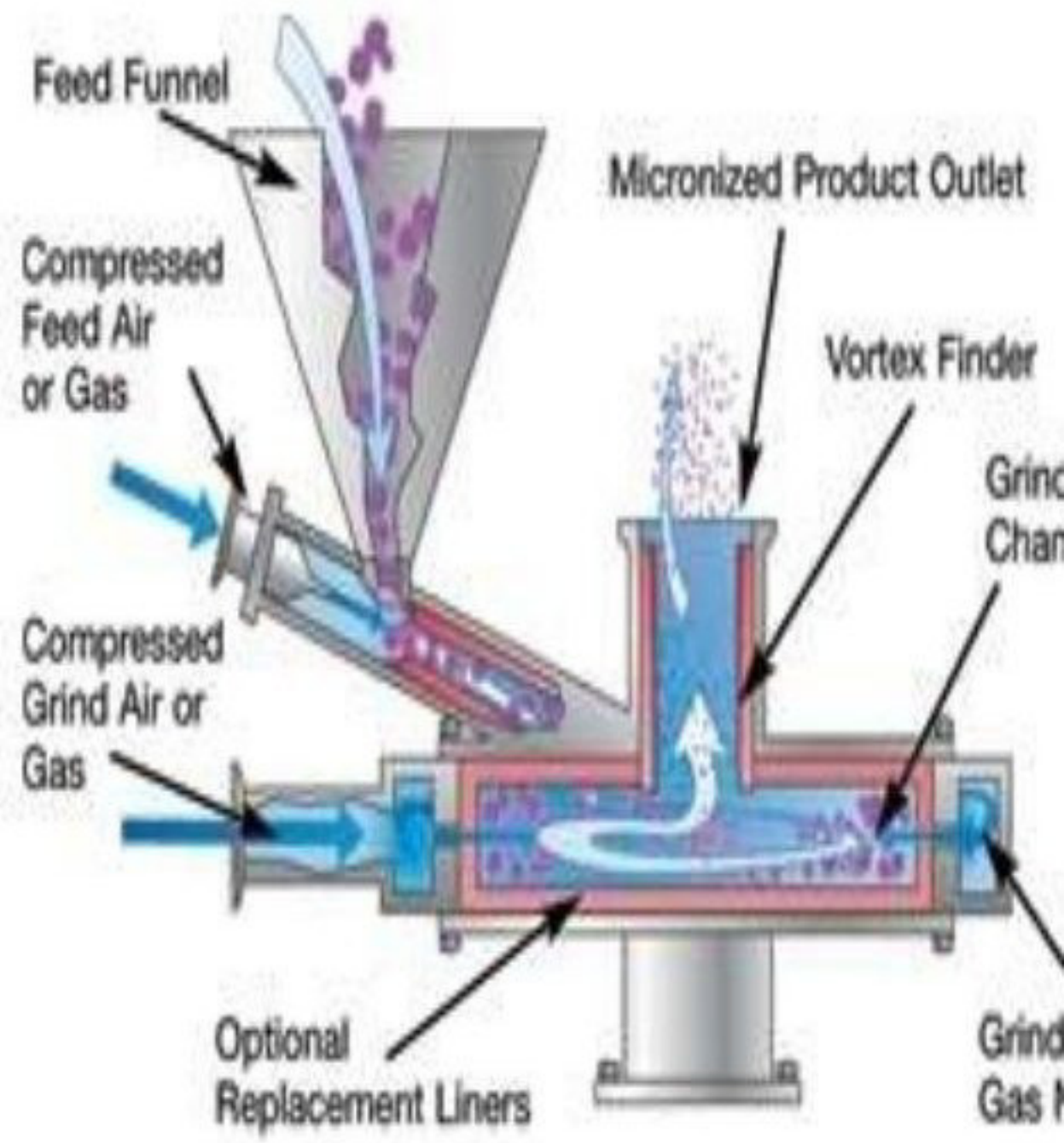
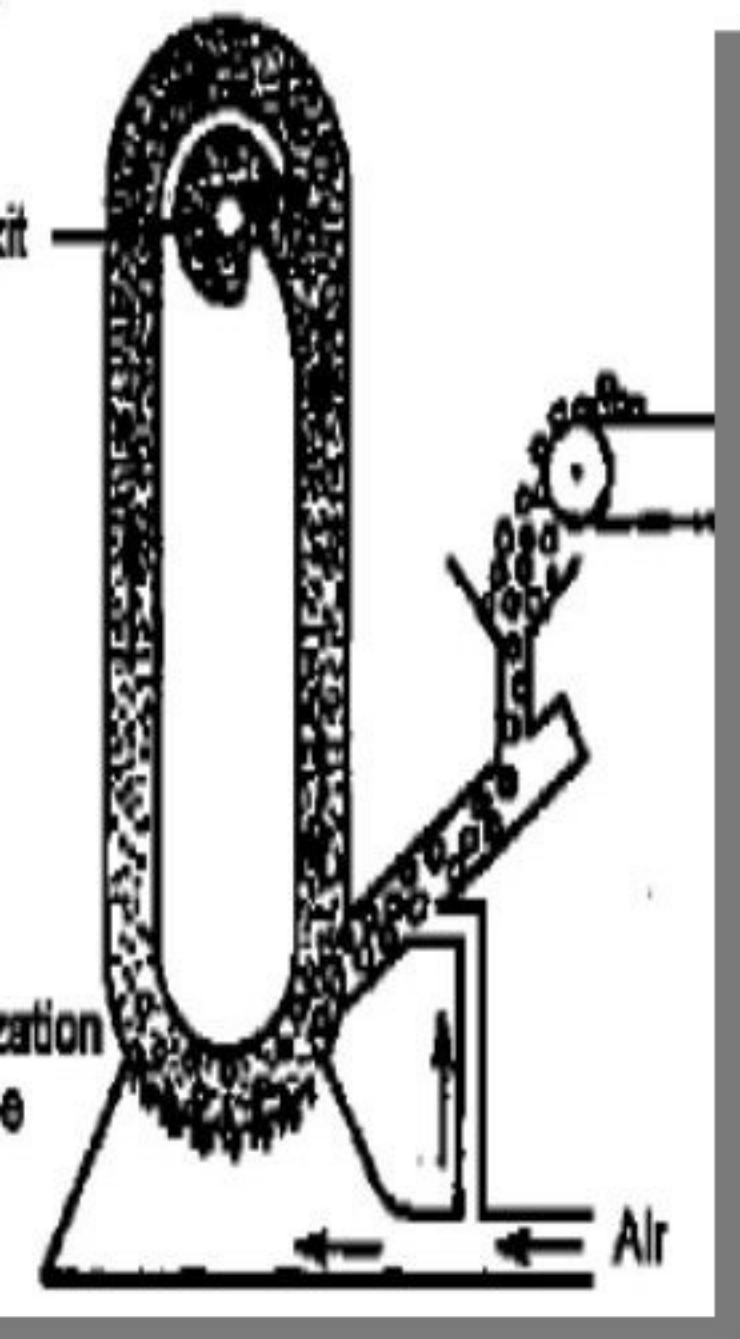
Working Mechanism



Industrial Model

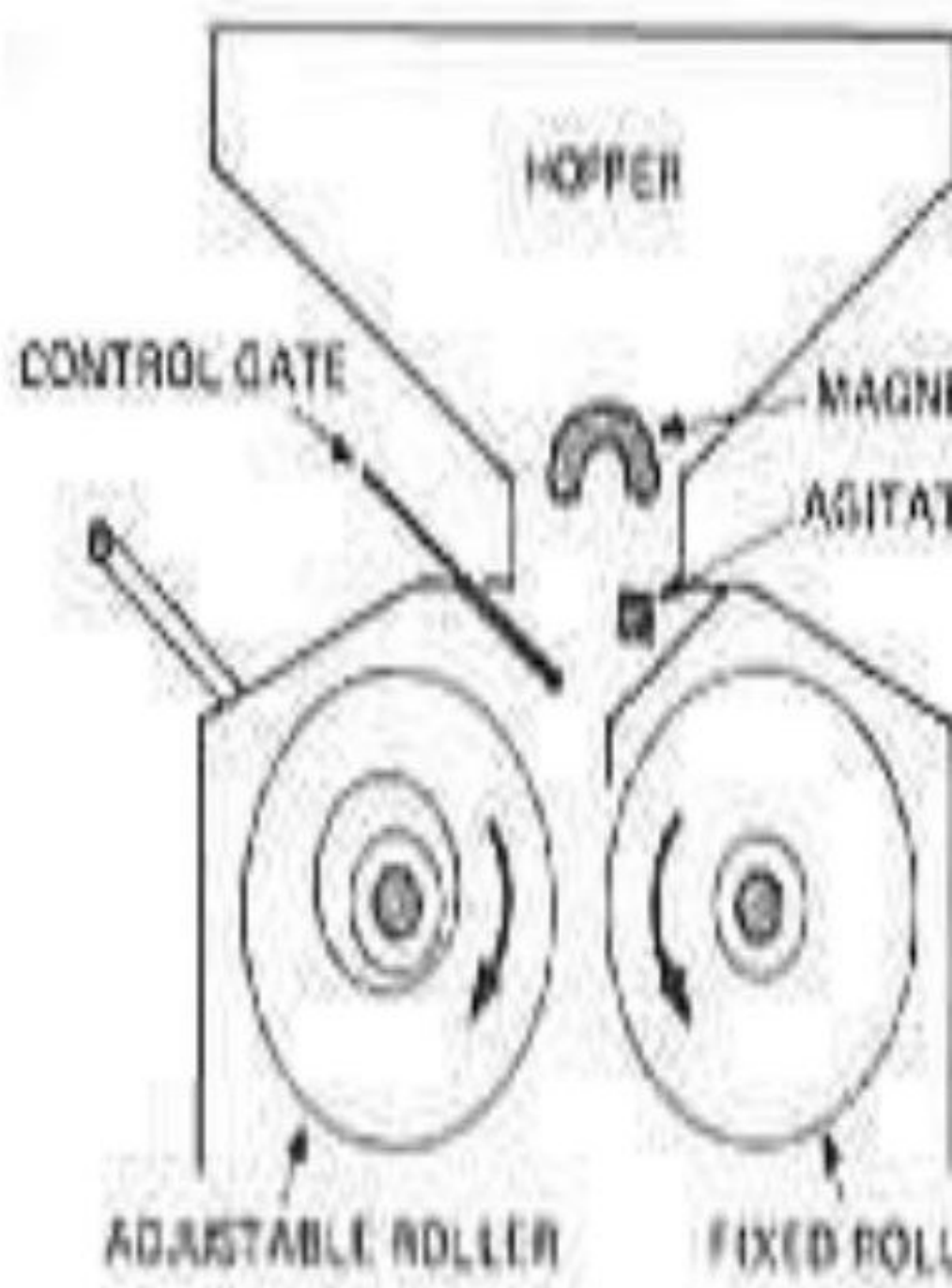


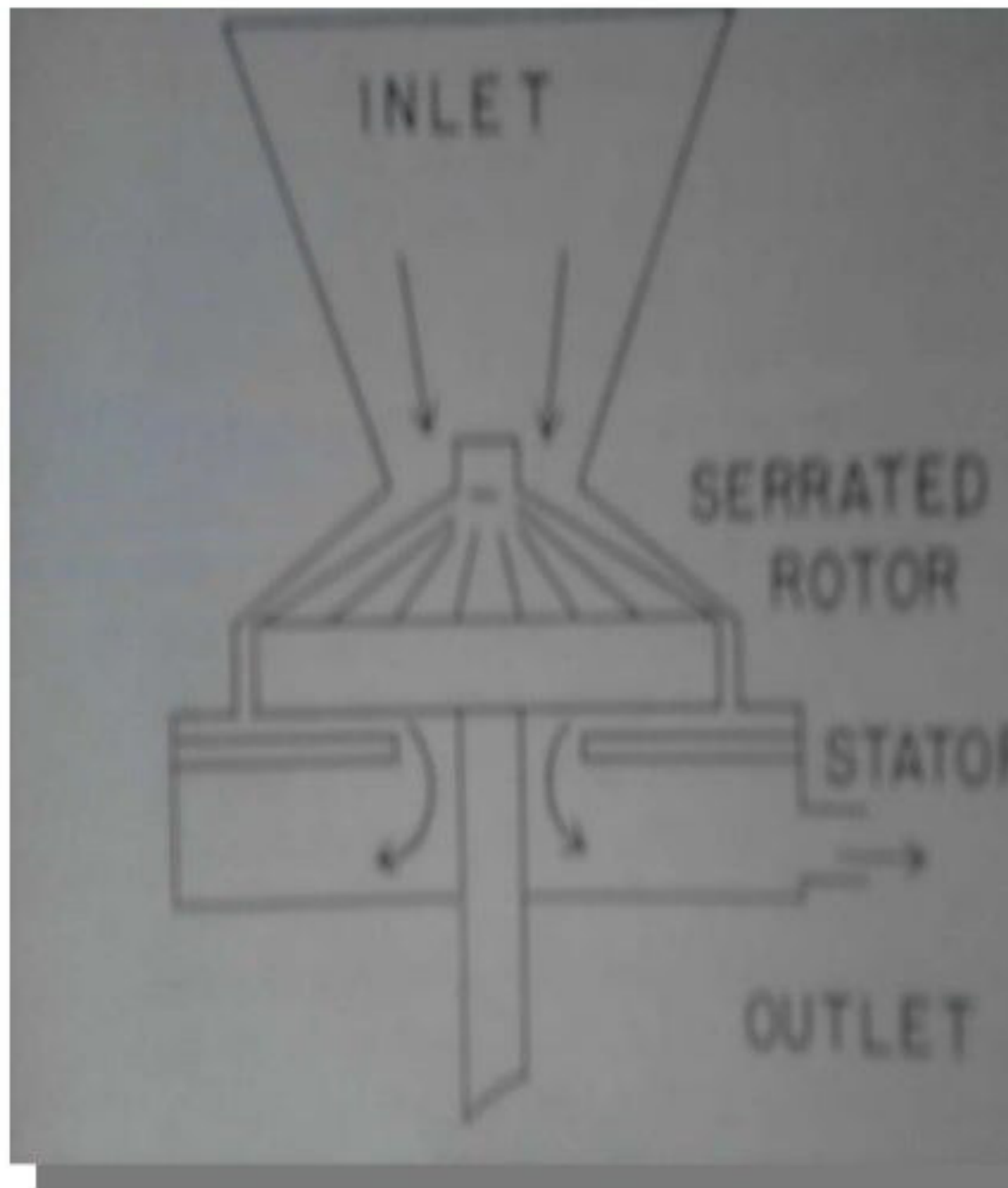
Fluid Energy Mill



Rolling with

Roller with





Tablet Production

Powders intended for compression into tablets must possess

two essential properties

Powder fluidity

- ✓ The material can be transported through the hopper into the die
- ✓ To produce tablets of a consistent weight
- ✓ Powder flow can be improved mechanically by the use of vibrators, incorporate the glidant

Powder compressibility

The property of forming a stable, intact compact mass when

Tableting Procedure

- **Filling**
- **Compression**
- **Ejection**

Tablet Compression Machines

Hopper for holding and feeding granulation to be compressed

Dies that define the size and shape of the tablet

Punches for compressing the granulation within the dies

Cam tracks for guiding the movement of punches

Feeding mechanisms for moving granulation from the hopper into the dies

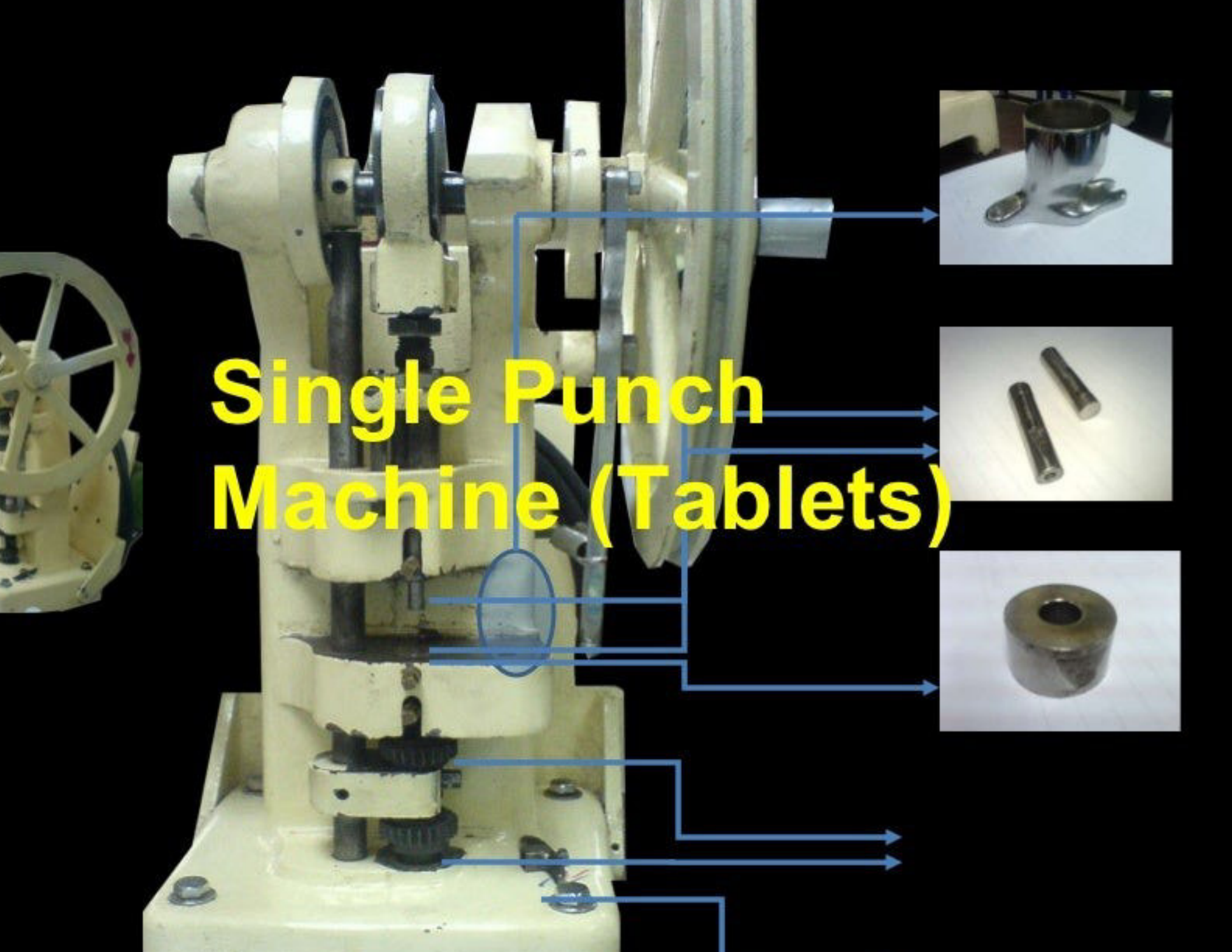
Single Punch Machine

The compression is applied by the upper punch

Stamping press



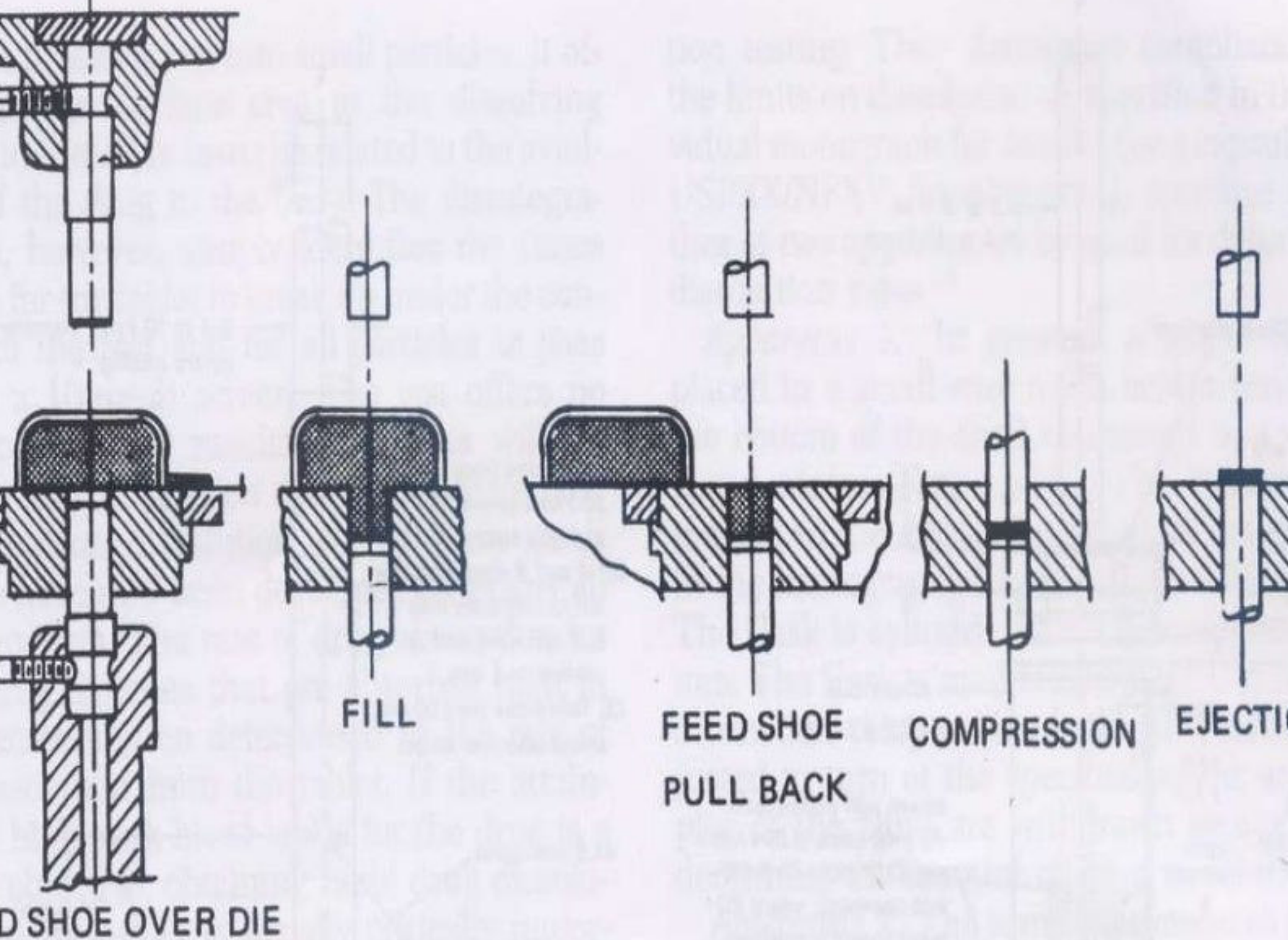
Single Punch Machine (Tablets)



top punch



bottom punch



5. The compression cycle of a single-munch tablet press (Courtesy of Vector Corporation, Marion, IA)

Multi-station Rotary Presses

The head of the tablet machine that holds the upper punches and lower punches in place rotates

As the head rotates, the punches are guided up and down by fixed **cam tracks**, which control the sequence of filling, compression and ejection.

The portions of the head that hold the upper and lower punches are called the **upper and lower turrets**

The portion holding the dies is called the **die table**

The **pull down cam (C)** guides the lower punches to the bottom, allowing the dies to overflow

The punches then pass over a **weight-control cam (E)**, which reduces the fill in the dies to the desired amount

A **swipe off blade (D)** at the end of the feed frame removes the excess granulation and directs it around the turret and into the front of the feed frame

while simultaneously the upper punches ride beneath the
upper compression roll (G)

The upper punches enter a fixed distance into the dies, while
the lower punches are raised to squeeze and compact the
granulation within the dies

After the moment of compression, the upper punches are
withdrawn as they follow the **upper punch raising cam (H)**

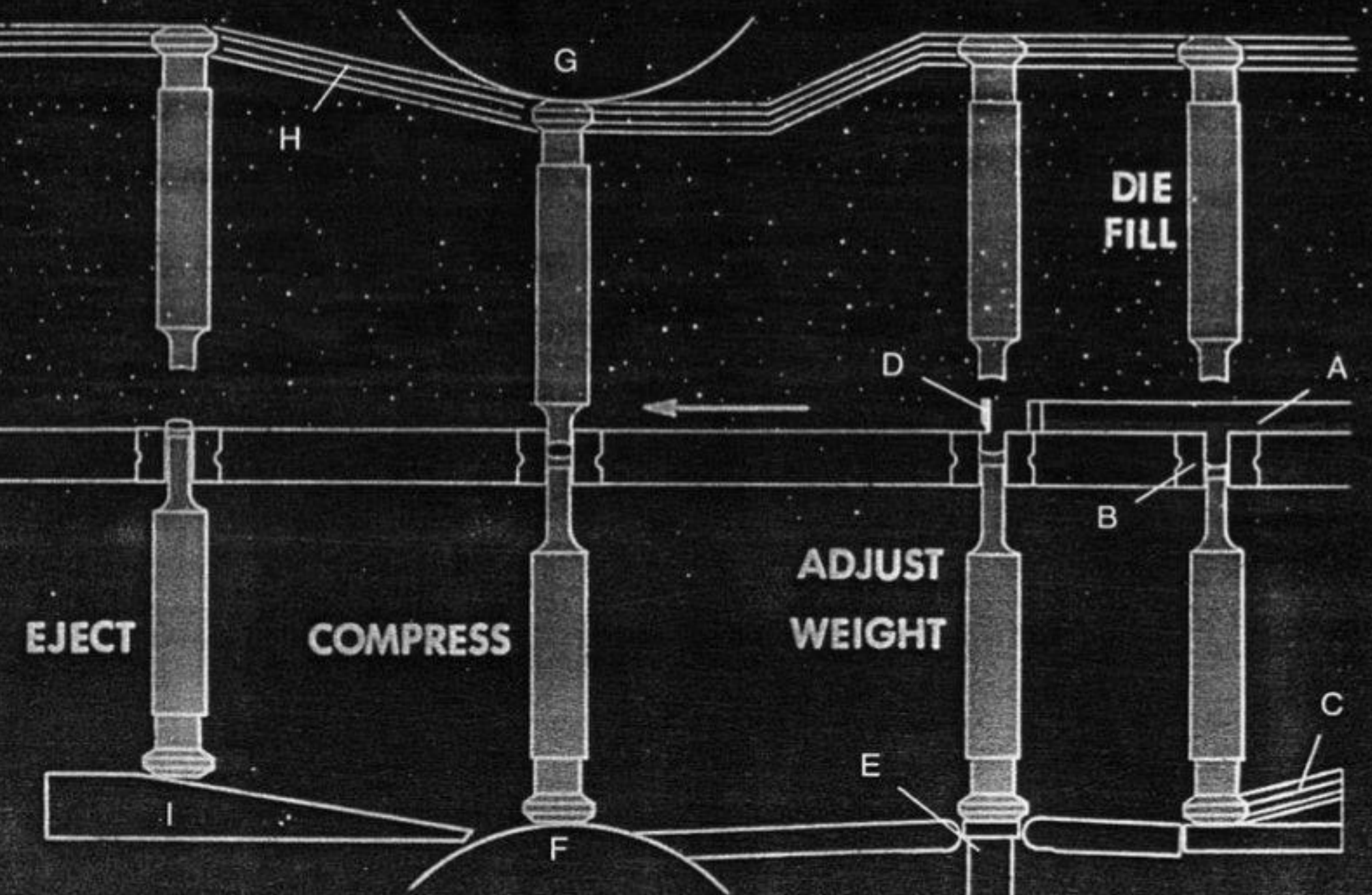
The lower punches ride up the **cam (I)** which brings the table

flush with or slightly above the surface of the dies

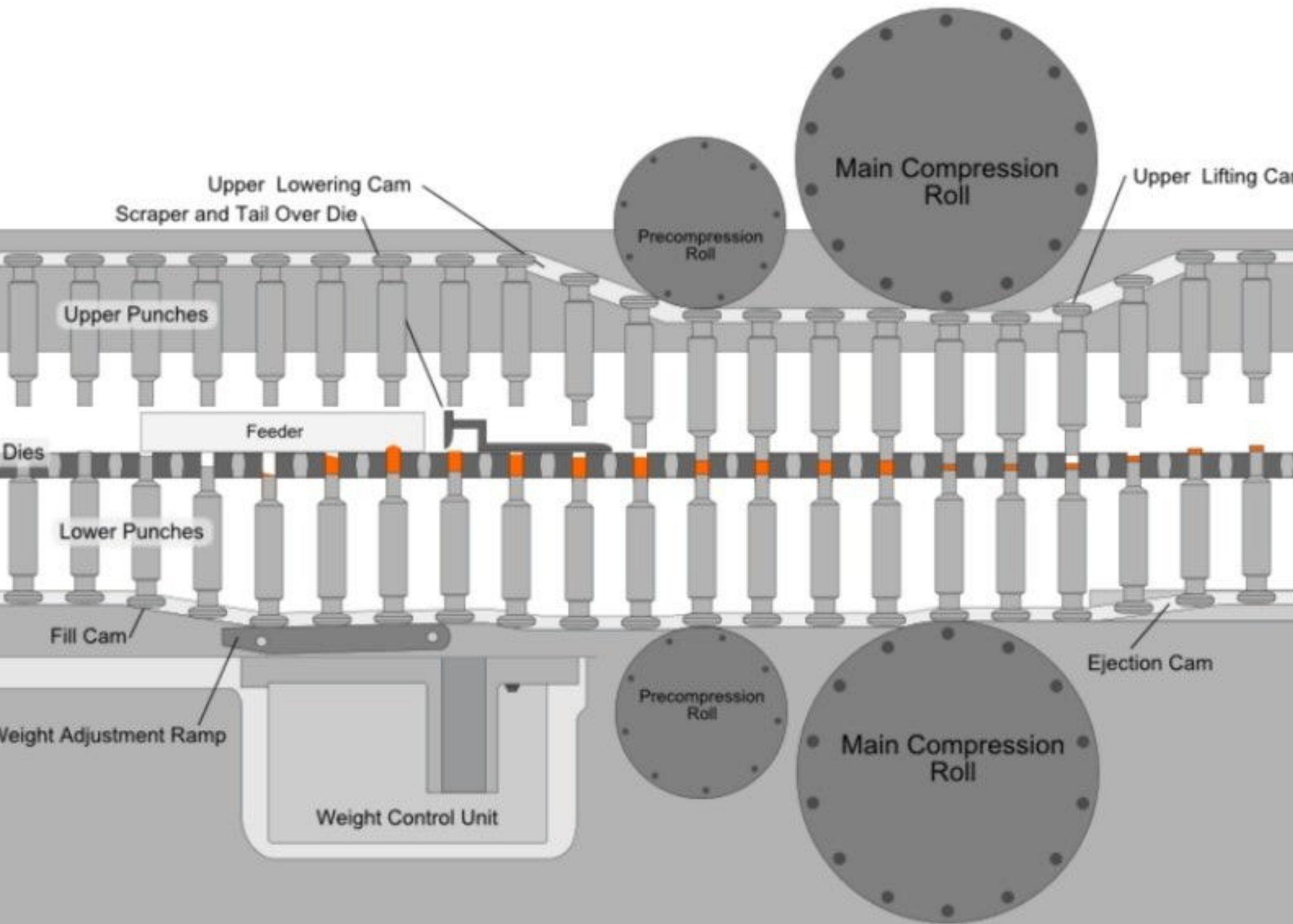
The tablets strike a sweep off blade affixed to the front of the **feed frame (A)** and slide down into a receptacle

At the same time, the lower punches re-enter the **pull down cam (C)** and the cycle is repeated





11-6. The compression cycle of a rotary tablet press. (See text for explanation of lettered labels.) (Courtesy of The American Hoffman-La Roche Co.)



through tablet compressing machinery has undergone numerous mechanical modifications over the years, the compaction of materials between a pair of moving punches within a stationary die has **remained unchanged**

The principle modification from earlier equipment has been an increase in production rate which is regulated by

- ✓ Number of tooling sets
- ✓ Number of compression stations
- ✓ Rotational speed of the press

Special adaptations of tablet machines allow for the compression of layered tablets and coated tablets

A device that chills the compression components to allow the compression of low-melting point substances such as waxes i.e. suppositories

Compression Machine Tooling

PUNCHES

- **BB Tooling:**
5.25" L, BD 0.75", 1".
- **B Tooling:** LP 3 ⁹/₁₆"
- **D Tooling:** Large tablets
L 5.25", BD 1", 1 ¹/₄" HD.

DIES

- OD 0.945" 7/16" RT or 9/16" Cap T
- OD 1 ³/₁₆" – 9/16" RT, 3/3" Cap T

Evaluation of Tablet

General Appearance:

The general appearance of a tablet, its identity and general elegance is an important factor for consumer acceptance, for control of lot-to-lot uniformity and tablet uniformity. The control of general appearance involves the measurement of size, shape, color, presence or absence of odor, taste etc.

Size & Shape:

Tablet size and shape can be dimensionally described & controlled. The thickness of a tablet is one of the important variables. Tablet thickness can be measured by micrometer or by other methods. Tablet thickness should be controlled within a $\pm 5\%$ variation of standard.

Unique identification marking:

These marking utilize some form of embossing, engraving or printing.

These markings include company name or symbol, product code, product name etc.

Organoleptic properties:

Color distribution must be uniform with no mottling. For visual color comparison compare the color of sample against standard color.

Hardness :

Tablet requires a certain amount of strength or hardness and resistance to friability to withstand mechanical shocks of handling in manufacture, packaging and shipping. Hardness generally measured

Different Hardness Tester



Monsanto



Pfizer



Strong-c

Friability:

Ability of a tablet can determine in laboratory by Roche friabilator. They consist of a plastic chamber that revolves at 25 rpm, dropping the tablets through a distance of six inches in the friabilator, which is then operated for 10 revolutions. The tablets are reweighed. Compress tablets that lose less than 1.0 % of the Tablet weight are considered acceptable.



7. Drug Content and Release:

Weight Variation test (U.S.P.):

Take 20 tablets and weigh individually. Calculate average weight and compare individual tablet weight to the average. The tablet passes the U.S.P. test if no more than 2 tablets are outside the percentage limit and if no tablet differs by more than 2 times the percentage limit.

Content Uniformity Test:

Randomly select 30 tablets. 10 of these are assayed individually. The tablet passes

if 9 of the 10 tablets must contain not less than 85% and not more than 115% of the labeled drug content and the 10th tablet may not contain less than 75% and more

than 125% of the labeled content. If the above conditions are not met, examine 20 tablets

Disintegration Test (U.S.P.):

S.P. device to test disintegration uses 6 glass tubes that are 3" long; and 10 mesh screen at the bottom end. To test for disintegration time, tablets are placed in each tube and the basket rack is positioned in a 1-L beaker containing simulated gastric fluid or simulated intestinal fluid at $37 \pm 2^{\circ}\text{C}$ such that the tablets remain 2.5 cm below the surface of liquid on their upward movement and at least 2.5 cm from the bottom of the beaker in their downward movement. The basket containing the tablets up and down through a distance of 5 cm at a frequency of 28 to 32 cycles per minute. Floating of the tablets can be prevented by using perforated plastic discs on each tablet.

According to the test the tablet must disintegrate and all particles must pass through a 10 mesh screen in the time specified. If any residue remains, it must have a certain diameter.

Integration time: Uncoated tablet: 5-30 minutes

Coated tablet: 1-2 hours

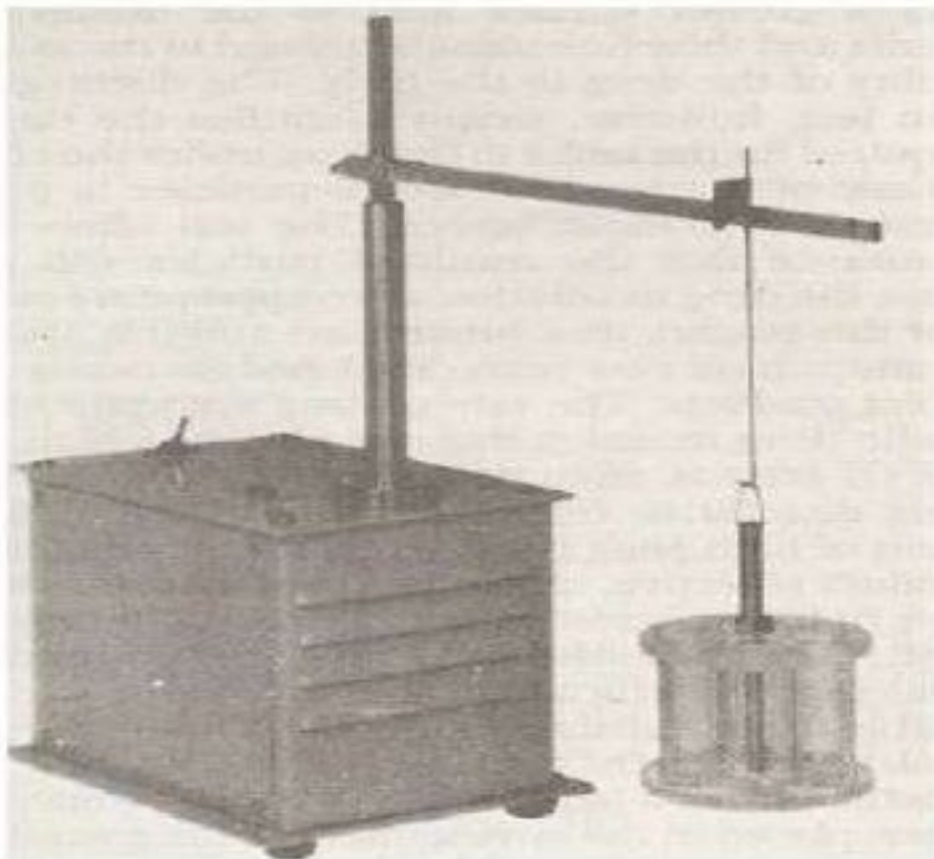


Fig. 19 Disintegration test apparatus



set of apparatus:

Apparatus-1:

Single tablet is placed in a small wire mesh basket attached to the bottom connected to a variable speed motor. The basket is immersed in a dissolution medium (as specified in monograph) contained in a 100 ml flask. The flask is cylindrical with a hemispherical bottom. The flask is maintained at $37 \pm 0.5^\circ\text{C}$ constant temperature bath. The motor is adjusted to turn at the specified rate. Samples of the fluid are withdrawn at intervals to determine the amount of drug in solutions.

Apparatus-2:

is same as apparatus-1, except the basket is replaced by a paddle. The dosage form is allowed to sink to the bottom of the flask before stirring. For dissolution test U.S.P. specifies the dissolution test medium and volume of apparatus to be used, rpm of the shaft, time limit of the test and any procedure for. The test tolerance is expressed as a % of the label amount of drug dissolved in the time limit.



Fig. 20 Dissolution test apparatus

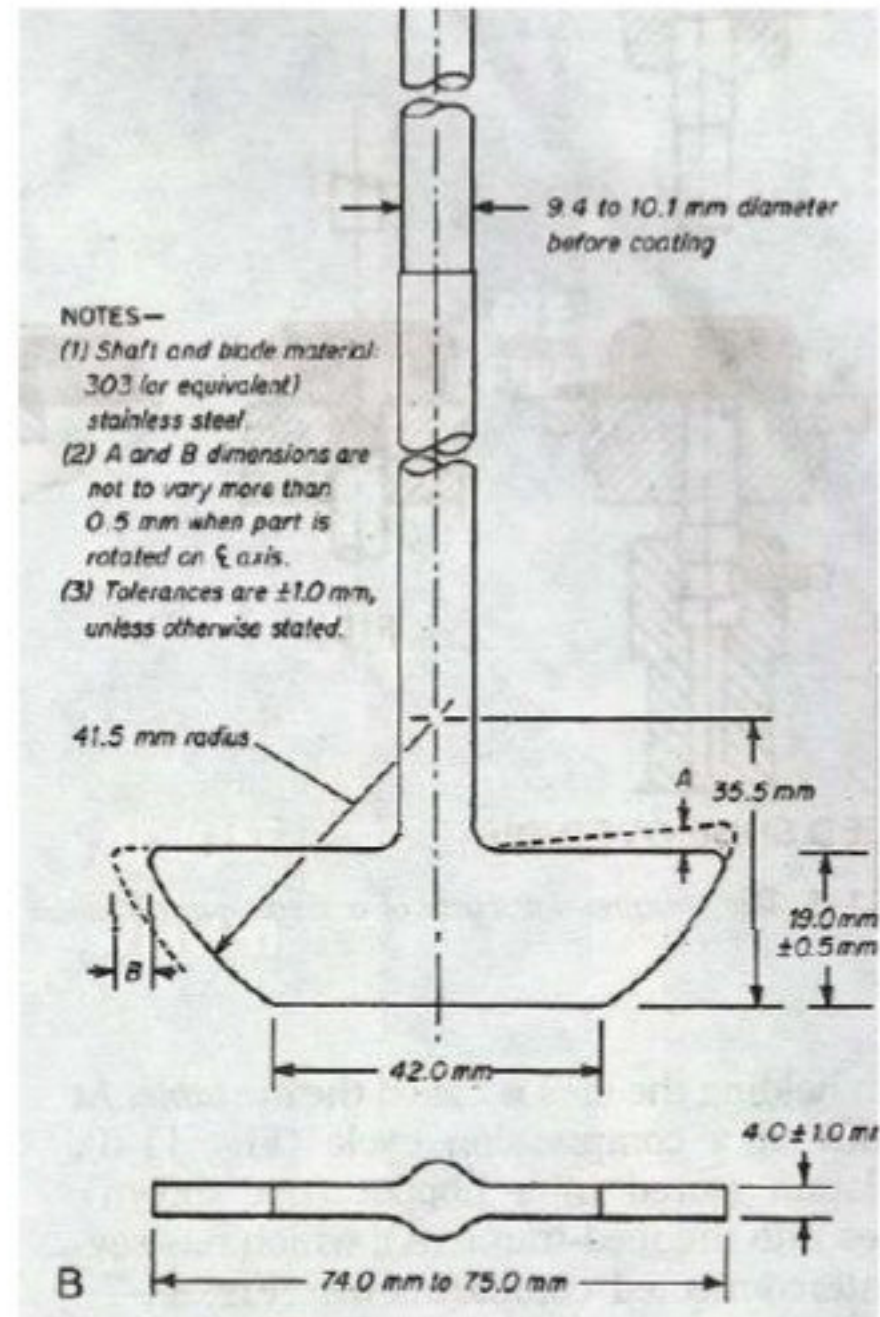


Fig. 22 USP dissolution apparatus 2

Problems In Tableting

Capping

Lamination / Laminating

Chipping

Cracking

Sticking / Filming

Picking

Binding

Mottling

Double impression

Granule Size and size distribution

Poor flow

Punch Variation

CAUSES	REMEDIES
Large amount of fines in the granulation	Remove some or all fines 100 to 200 mesh screen
Too dry or very low moisture content (leading to loss of proper binding action).	Moisten the granules suitable hygroscopic substance e.g. methyl-cellulose or PEG
Not thoroughly dried granules.	Dry the granules properly
Insufficient amount of binder or improper binder.	Increasing amount of binder Adding dry binder such gelatinized starch, gum powdered sorbitol, PVP, hydroxy silica or powdered sucrose
Insufficient or improper lubricant.	Increase the amount of lubricant change the type of lubricant
Granular mass too cold to compress firm.	Compress at room temperature

No.	CAUSES	REMEDIES
	Poorly finished dies	Polish dies properly. Investigate steels or other materials.
	Deep concave punches or beveled-edge faces of punches.	Deep concave punches or beveled faces of punches.
	Lower punch remains below the face of die during ejection.	Make proper setting of lower during ejection.
	Incorrect adjustment of sweep-off blade.	Adjust sweep-off blade correctly to facilitate proper ejection.
	High speed	Reduce speed of turret (Increase)

CAUSES	REMEDIES
Rapid relaxation of the peripheral regions of a tablet, on ejection from a die.	Use tapered dies,
Rapid decompression in the pre-compression step.	Reduce turret speed and reduce compression pressure.

Causes And Remedies Of Chipping Related To Formulation (Granulation) Are As Follows

No.	CAUSES	REMEDIES
1.	Sticking on punch faces	Dry the granules properly or increase lubrication.
2.	Too dry granules.	Moisten the granules to plasticize. Add hygroscopic substances.
3.	Excessive lubrication	

MACHINE (DIES, PUNCHES AND TABLET PRESS)

CAUSES	REMEDIES
Groove of die worn at compression point.	Polish to open end, reverse or replace the die
Barreled die (center of the die wider than ends)	Polish the die to make it cylindrical
Edge of punch face turned inside/inward.	Polish the punch edges
Concavity too deep to compress properly.	Reduce concavity of punch faces. Use flat punch

CAUSES AND REMEDIES OF CRACKING RELATED TO FORMULATION (GRANULATION)

CAUSES	REMEDIES
Large size of granules.	Reduce granule size. Add fines.
Too dry granules.	Moisten the granules properly and add proper amount of binder
Tablets expand.	Improve granulation. Add dry binders
Granulation too small.	Improve granulation. Add dry binders

MACHINE (DIES, PUNCHES AND TABLET PRESS)

CAUSES	REMEDIES
Tablet expands on ejection due to air entrapment.	Use tapered die.
Deep concavities cause cracking while	removing tablets Use special take-off

THE CAUSES AND REMEDIES OF STICKING RELATED TO FORMULATION (GRANULATION)

CAUSES	REMEDIES
Granules not dried properly.	Dry the granules properly. Make moisture analysis to determine limits.
Too little or improper lubrication	Increase or change lubricant.
Too much binder	Reduce the amount of binder or use different type of binder.
Hygroscopic granular material.	Modify granulation and compress under controlled humidity.
Oily materials	Modify mixing process. Add an absorbent.

CAUSES**REMEDIES**

Concavity too deep for granulation.

Reduce concavity to optimum.

Too little pressure.

Increase pressure.

Compressing too fast.

Reduce speed

CAUSES AND REMEDIES OF PICKING RELATED TO FORMULATION (GRANULATION)**CAUSES****REMEDIES**

Excessive moisture in granules.

Dry properly the granules, determine optimum limit.

Too little or improper lubrication.

Increase lubrication; use colloidal silica as 'polishing agent', so that material does not cling to punch faces.

Low melting point substances, may soften from the heat of compression and lead to picking.

Add high melting-point materials. Use high melting-point lubricants.

Low melting point medicament in high concentration.

Refrigerate granules and the entire tablet press.

Too warm granules when compressing

Compress at room temperature. Cool sufficiently before use.

CAUSES	REMEDIES
Rough or scratched punch faces.	Polish faces to high luster.
Embossing or engraving letters on punch faces such as B, A, O, R, P, Q, G.	Design lettering as large as possible. Plate the punch faces with chromium to give a smooth and non-adherent face.
Pressure applied is not enough; too soft tablets.	Increase pressure to optimum.

CAUSES AND REMEDIES OF BINDING RELATED TO FORMULATION (GRANULATION)

CAUSES	REMEDIES
Too moist granules and extrudes around lower punch.	Dry the granules properly.
Insufficient or improper lubricant.	Increase the amount of lubricant or use a more effective lubricant.
Too coarse granules.	Reduce granular size, add more fines, and increase quantity of lubricant.
Too hard granules for the lubricant to be effective.	Modify granulation. Reduce granular size.
Granular material very abrasive and cutting into dies.	If coarse granules, reduce its size. Use wear-resistant dies.

CAUSES	REMEDIES
Poorly finished dies.	Polish the dies properly.
Rough dies due to abrasion, corrosion.	Investigate other steels or other materials or granulation.
Undersized dies. Too little clearance.	Rework to proper size. Increase clearance.
Too much pressure in the tablet press.	Reduce pressure. OR Modify granulation.

CAUSES AND REMEDIES OF MOTTLING

CAUSES	REMEDIES
A coloured drug used along with colourless or white-coloured excipients.	Use appropriate colourants.
A dye migrates to the surface of granulation while drying.	Change the solvent system, Change the binder, Reduce drying temperature and Use a smaller particle size.
Improperly mixed dye, especially during 'Direct Compression'.	Mix properly and reduce size if it is of a larger prevent segregation.
Improper mixing of a coloured binder solution.	Incorporate dry colour additive during powder step, then add fine powdered adhesives such as ac

Tablet Coating

Tablet coating objectives.

Three Primary components involved in tablet coating

Tablet properties

Coating process

- Coating equipment.
- Parameters of the coating process.
- Facility and ancillary equipment.
- Automation in coating processes.

Coating compositions

1. Tablet Properties

- Mechanical and Physical Strength
- Smooth surface
- Physical shape
- Chemical nature of tablet ingredients
- Hygroscopicity

- **Three types of equipments**
 1. The standard coating pan
 2. The perforated coating pan
 3. The fluidized bed (Air suspension) coater.

- **These systems based on three basic designs**
 - 1. Conventional pan system**

Depending on drying efficiency

 - Pellegrini system
 - Immersion-sword system
 - Immersion –tube system
 - 2. Perforated pan system**
 - Accela-coata
 - Hi-coater systems
 - Driacoater
 - Glatt coater
 - 3. Fluidized bed (Air suspension) system**

inches diameter revolving on its horizontal axis.

ed air is directed **into** the **pan** and **onto** the **tablet bed**

e.

usted by means of **ducts** positioned through the **front of**

n

ing efficiency is achieved by,

legrini pan:

s a **baffled pan** and

user for uniform distribution

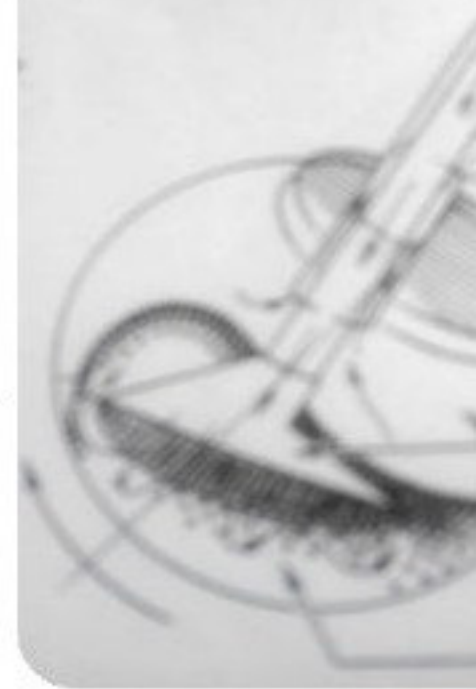
rying air.

closed and **automated**.



ing air through perforated
sword immersed in the
bed.

ward flow through bed.
ay onto the bed surface



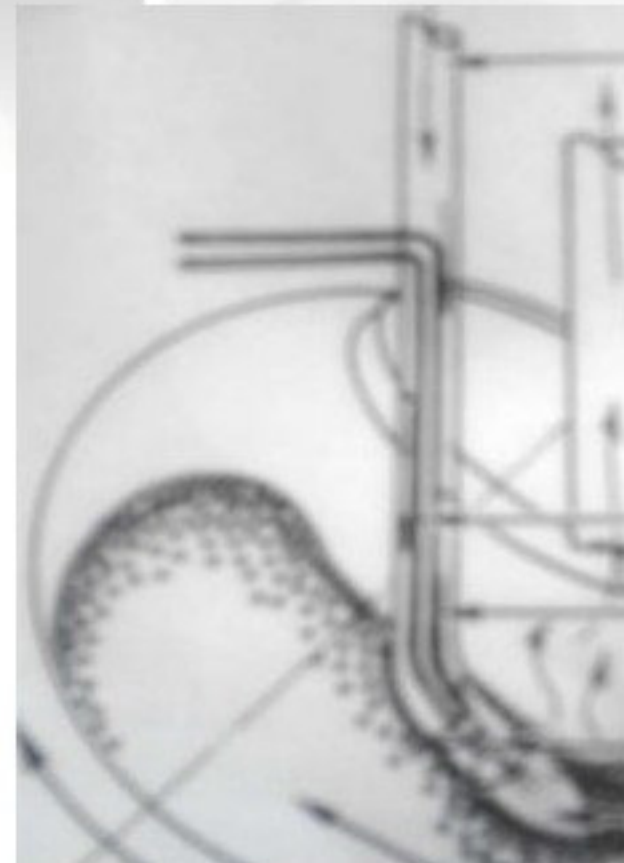
ersion-tube system

e immersed in tablet bed

ers heated air and coating solution through

y built in the tip of tube

s upward and exhausted by a conventional duct.



ated drum rotates on its horizontal axis in an
ed housing.

Hi-Coata and **Hi-coater system**: Drying air is
ed in to drum, is passed through bed, and is
sted through perforations in to drum.



Eight Gun Manifold Spray Bar
Titanium or stainless steel construction available.



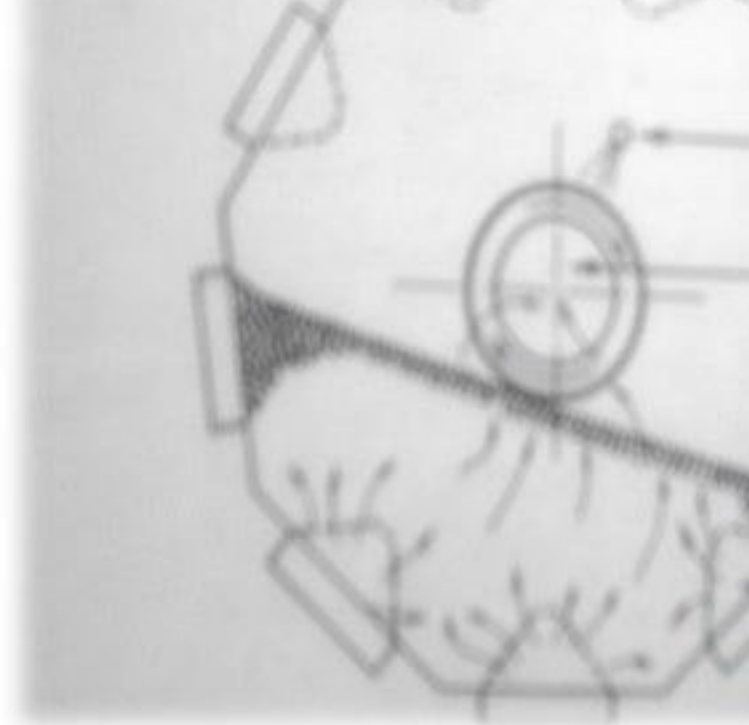
Water:

roduces **drying air** through hollow **perforated**
s located inside periphery of the drum.

n rotating pan, **ribs dip into bed**

ing air **passes up** through and **fluidizes bed**

haust is from **back** of the **pan**



coater:

ing air directed **from inside** the drum

ugh **bed** and out an **exhaust duct**

a an optional split-chambered plenum

ng air can be directed in the **reverse** manner

eral air flow configurations are possible.

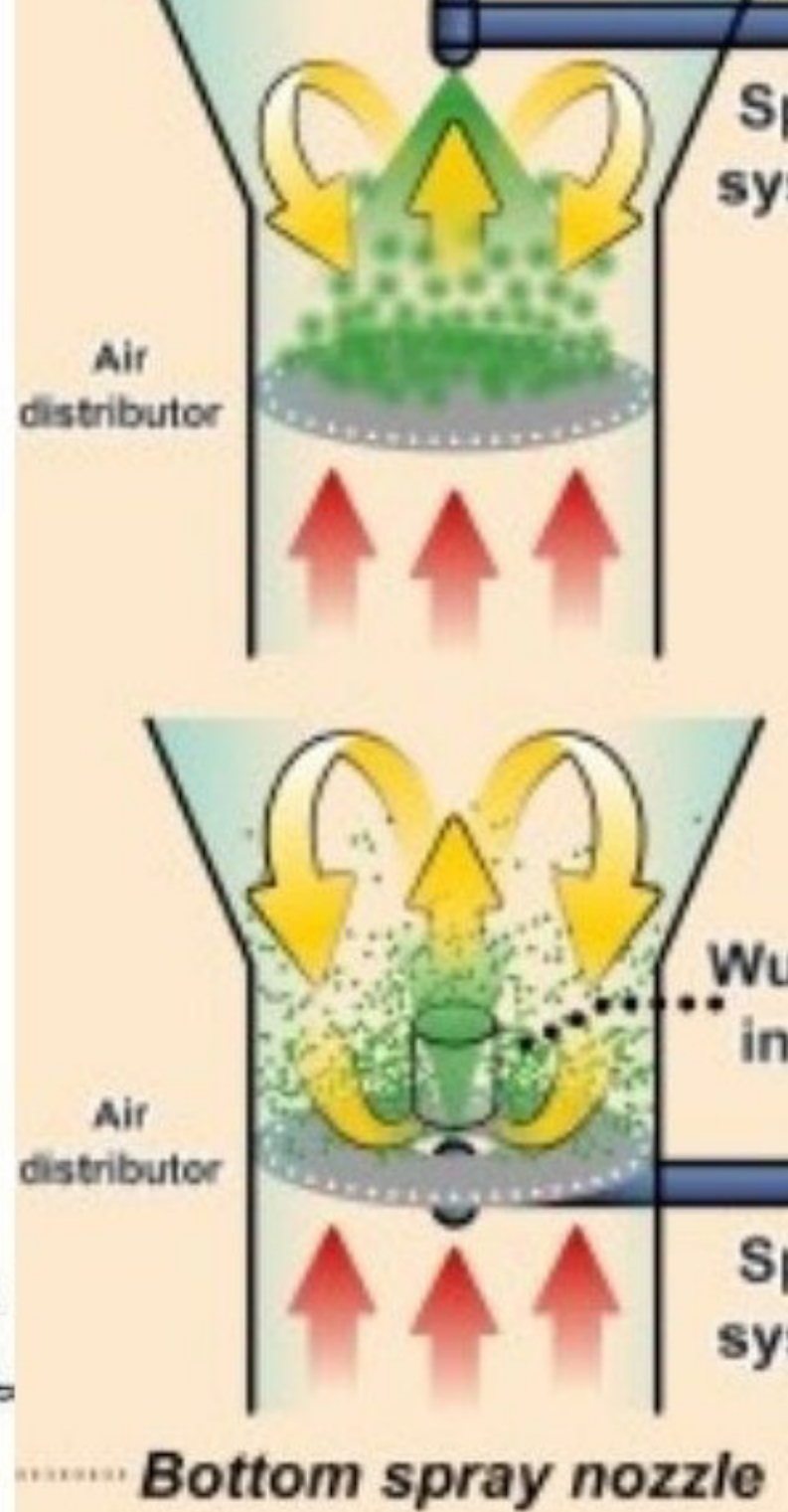
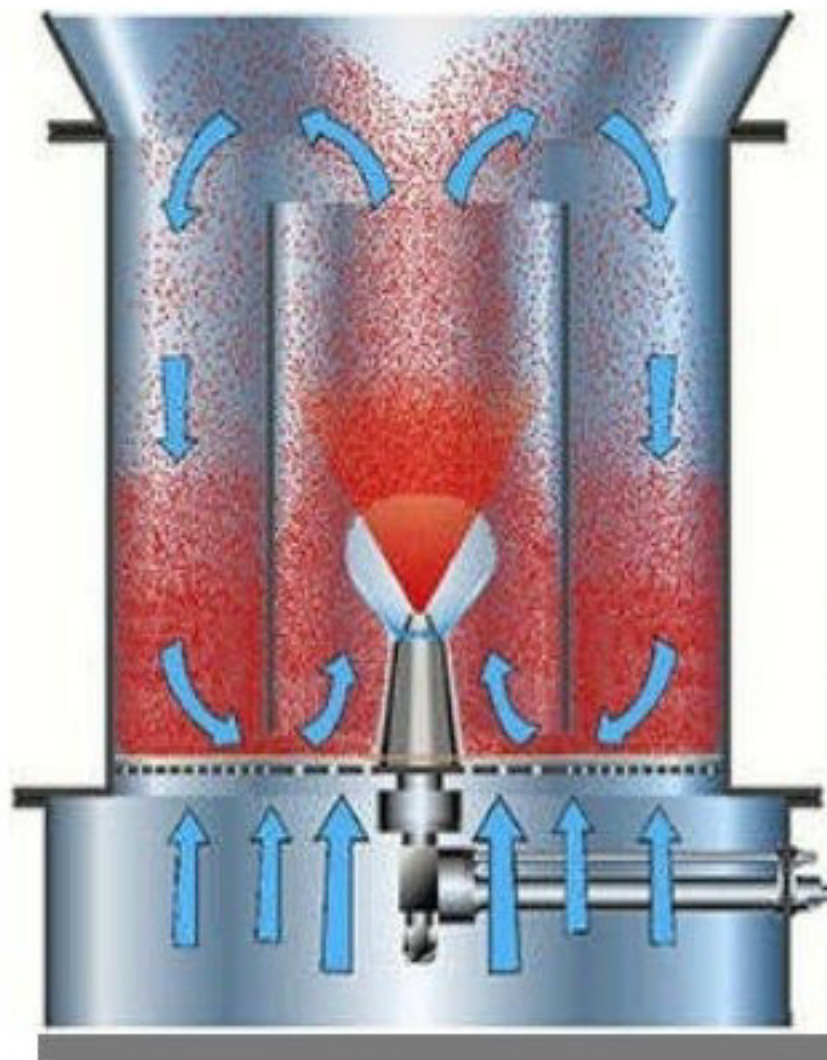


highly efficient

fluidization of tablet bed is achieved in a

cylindrical chamber by the upward flow of

heating air.



fluid bed

Spray application system

Basic **types** of **spray system** differ in manner in which **atomization** of liquid is achieved

High-pressure, airless (a)

High pressure liquid (250-3000psig) through a small orifice (.009" - .020")

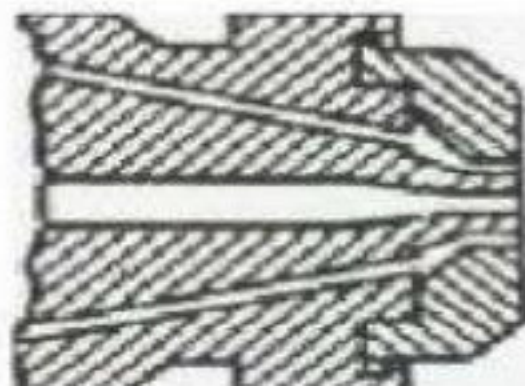
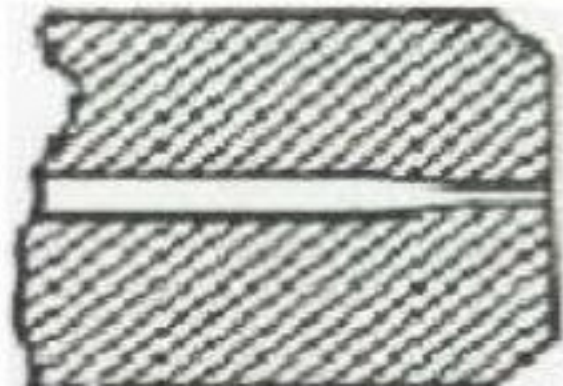
Degree of atomization- fluid pressure, orifice size, viscosity.

Low-pressure, air-atomized (b)

Low pressure liquid (5-50psig) through a large orifice (.020" - .060" id)

Degree of atomization- fluid pressure, orifice size, viscosity, air pressure

air cap design.



Rate of coating composition application = Rate of evaporation of volatile solvent

Deviation from this results a serious coating problems

Mathematical modeling for automated aqueous coating process



re,

H) = Air capacity,

$C(S)$ = Coating Composition

SA = Tablet surface area,

E = Equipment efficiency

Capacity:

Quantity of water or solvent removed during coating process

- depends on :
1. Quantity of air flow through tablet bed,
 2. Temperature of air,
 3. water content of inlet air.

Drying Composition:

inlet air-heated, exhaust air- cool.

Thickness of film- rate of application

For rapidly drying formulations dry quickly on the tablet surface allowing

constant application by efficient atomization of coating solution

and presence of debossed features affects coating conditions

total surface area per unit weight decreases from smaller to larger tablets

for same thickness of film, smaller tablets requires more coating composition

compared with larger tablets

size of atomized coating droplets must be smaller and better controlled as

features to be coated become smaller.

Coating efficiency, E:

Net increase in coated tablet weight

Coating Efficiency, E = -----

Total nonvolatile coating weight applied to tablets

Facility and Ancillary Equipment

Facility should meet to requirement of cGMP

Adequate space for equipment, processing, in-process storage

Special requirements depending on nature of solvent- electrical explosion-pro

Specialized ventilation

Best air treatment to recover solvent or to prevent entry in to atmosphere

Federal EPA defines limits of organic solvents and particulate allowed in atm

Water based coating is advantageous

Equipments

Grinders, mixers, mills, jacketed tanks, portable pressure tanks or pumping

Tablet Coating Process

Final production step on which quality of product may be judged

Sugar Coating

involves,

1. Sealing,
2. Subcoating
3. Syruping (Smoothing)
4. Finishing
5. Polishing

prevent moisture penetration into tablet core

especially in pan-ladling process in which localized overwetting of portion of

tablet bed occur

cellac

affects disintegration and dissolution time due to polymerization on aging



n

such effect is not reported

coating

around the edges and build up size .

coating Steps

binder solution  Dusting of subcoating powders  Dryin

Final Coating

cover and fill in the imperfections in tablet surface caused by subcoating

impart desired color

most syrup coat contains some suspended powders i.e. glossing syrups

white colorants— tinted base— uniform coating

syrup solution containing dye applied until final size and color are achieved

Polishing

final syrup coating step

few clear coats of syrup may be applied

Polishing

desired luster is obtained in this final step

can standard coating pan, canvas-lined coating pans

Coating are given below:

Coating	Sub coating	Syrup coating	Polishing sol
Shellac	Gelatin	Colorant	Carnauba wa
Acid	Acacia	Sub coating powder	(yellow)
Ethylene glycol	Sugar cane powder	Cal. Carbonate	Bees wax
1000	Corn syrup	Cane sugar powder	(white)
Ethylene chloride	Syrup	Corn starch	Paraffin wax
1	Distilled water	Syrup	Naphtha
		Distilled water	

Coating polymers: Cellulose acetate phthalate, Acrylate poly
propyl methyl cellulose phthalate, Polyvinyl acetate phthalate

Solvents used for coating: Ethanol, Methanol, Isopropanol, Chloroform, Aceton
ethylene chloride, Methylene ethyl ketone

Film Coating

film coating and sugar coating shares the same equipments and process

parameters

two methods,

1) Pan-Pour method

is similar to that of pan-pour sugar coating

but it is relatively slow and relies heavily on skill and technique of operator

continuous based film coating is not suitable due to localized over-wetting

2) Pan-Spray method

uses an automated spraying system

ensure the consistent product quality, certain elements of process need to be controlled regardless of coating pan system

The variables to be controlled in pan-spray film coating process are:

Pan variables

- Pan design/ baffling
- Speed
- Pan load

Process air

- Air quality
- Temperature
- Airflow rate/ volume/ balance

Spray variables

- Spray rate
- Degree of atomization
- Spray pattern

er vapor permeability

tensile strength

ted Tablet Evaluation

Adhesion test with tensile strength tester

Diametral crushing strength of coated tablets Rate of coated tablet

disintegration/ dissolution

Stability studies

Surface roughness, hardness, color uniformity through instrumental me

Visual inspection for coated tablet quality.

Qualitative measure of resistance of a coated tablet to abrasion by whi

Coating Formula Optimization

Modifications in basic formula,

- To improve adhesion of the coating to the core
- To decrease bridging of inclusions
- To increase coating hardness
- To improve any other property as per need of formulation

Common modifications are

- Changes in polymers-to-plasticizers ratio
- Addition of different polymers or plasticizers

and have following attributes;

solubility in solvent

solubility required for intended use

capacity to produce elegant looking product

stability with heat, light, moisture, air, substrate. Essentially no color, taste

compatibility with common coating solution additives

non-toxic, therapeutically inert and ease of application to the particles or tablet

resistance to cracking, and provision of adequate moisture, light, odor, or

protection barrier when desired

bridging or filling of the debossed tablet surfaces by the film former

Film Formers

classified in *Nonenteric* and *Enteric* Materials

Nonenteric Materials

- HPMC,
- MHEC,
- EC,
- HPC,
- Povidone,
- SCMC,
- PEG,
- Acrylate Polymers

Reasons for enteric coating

Protect acid labile drugs from gastric fluid e.g. enzymes, ATBTs

Prevent gastric distress or nausea due to irritation from drug, e.g. Sodium

Deliver drugs for local action in intestines, e.g. Intestinal antiseptic (Kant)

Drugs optimally absorbed in small intestine

Provide delayed release component for repeat-action tablets

Material should have following properties

Resistance to gastric fluids

Ready susceptibility or permeability to intestinal fluids

Stability with coating composition and drug substrate

Stability alone and in coating solution

Formation of continuous film

Nontoxicity, Low cost

Free from interaction with other medicinal components

CAP,

Acrylate Polymers

HPMCP

PVAP

Solvents

Plasticizers

“**Internal**” or “**External**” techniques

Colorants

Opacuant-Extenders

Miscellaneous Coating Solution Components

- Quality Control**
- Stability Testing**
- Film Defects**
- Specialized Coatings**
 - **Compression Coating**
 - **Electrostatic Coating**
 - **Dip Coating**
 - **Vacuum Film Coating**

DEFECTS AND REMEDIES FOR TABLET COATING

- Blistering
- Chipping
- Cratering
- Picking
- Pitting
- Blooming
- Blushing
- Colour variation
- Infilling
- Orange peel/Roughness

• Cracking (Cracks)

CAUSES	REMEDIES
Effect of temperature on the strength, elasticity and adhesion of the film	Use mild drying condition.
THE CAUSE AND REMEDY OF CHIPPING	
High degree of attrition associated with the coating process.	Increase hardness of the film by increasing the molecular weight grade of polymer.
THE CAUSES AND REMEDIES OF CRATERING	
Inefficient drying.	Use efficient and optimum drying conditions.
Higher rate of application of coating solution.	Increase viscosity of coating solution to decrease application rate.
THE CAUSES AND REMEDIES OF PICKING	
Inefficient drying.	Use optimum and efficient drying conditions or increase inlet air temperature.
Higher rate of application of coating solution	Decrease the rate of application of coating solution or increasing viscosity of coating solution.
THE CAUSE AND REMEDY OF PITTING	
Inappropriate drying (inlet air) temperature	Dispensing with preheating procedures at the inlet of coating and modifying the drying (inlet air) temperature so that the temperature of the tablet core is not greater than the melting point of the batch of additives used.
THE CAUSE AND REMEDY OF BLOOMING	

	High coating temperature	Decrease the drying air temperature
	Use of sorbitol in formulation which causes largest fall in the thermal gelation temperature of the Hydroxy Propyl Cellulose, Hydroxy Propyl Methyl Cellulose, Methyl Cellulose and Cellulose ethers.	Avoid use of sorbitol with Hydroxy Propyl Cellulose, Propyl Methyl Cellulose, Methyl Cellulose and Cellulose ethers.

THE CAUSE AND REMEDY OF COLOUR VARIATION

	Improper mixing, uneven spray pattern, insufficient coating, migration of soluble dyes-plasticizers and other additives during drying.	Go for geometric mixing, reformulation with plasticizers and additives or use mild drying conditions.
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THE CAUSE AND REMEDY OF INFILLING

	Bubble or foam formation because of air spraying of a polymer solution	Add alcohol or use spray nozzle capable of atomization.
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THE CAUSES AND REMEDIES OF ORANGE PEEL/ROUGHNESS

	Rapid Drying	Use mild drying conditions
	High solution viscosity	Use additional solvents to decrease viscosity

THE CAUSE AND REMEDY OF CRACKING/SPLITTING

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A tropical sunset scene with palm trees and a body of water. The sky is a mix of orange, yellow, and red, with the sun low on the horizon. Two palm trees are visible, one on the left and one on the right, their fronds silhouetted against the bright sky. The water in the foreground is calm, reflecting the colors of the sunset.

Thank You

*If you salute your duties, No need to salute anybody
If you don't salute your duties, You have to salute everybody.*